

REGIOSELECTIVE INTRODUCTION OF ELECTROPHILES INTO PIPERIDINE DERIVATIVES AT THE 4-POSITION[†]

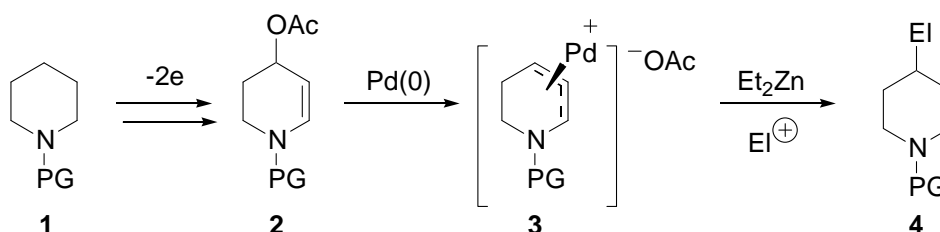
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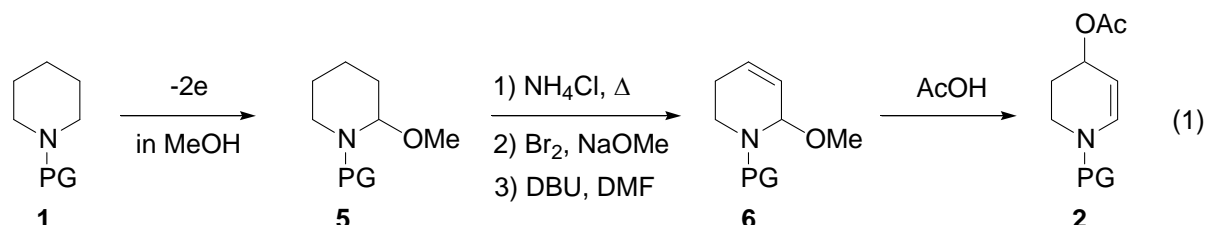
Abstract – Regioselective introduction of various electrophiles (aldehydes, ketones, and imines) into piperidine skeleton at the 4-position was achieved with a catalytic amount of Pd(OAc)₂/PPh₃ in the presence of excess Et₂Zn. In addition, enantioselective introduction of benzaldehyde into piperidine derivatives was accomplished by using chiral phosphine ligand with moderate enantioselectivity.

Piperidines possessing substituents at the 4-position are useful synthetic intermediates for a variety of natural products and drug candidates.¹ Accordingly, it is worthwhile to develop convenient methods for introduction of substituents at the 4-position of piperidine skeleton. Although some methods for the nucleophilic substitution are known,² the electrophilic substitution has not been reported to date. We wish to report herein regioselective introduction of various electrophiles (aldehydes, ketones, and imines) into piperidine derivatives at the 4-position. Our strategy for generation of nucleophilic species from piperidine derivatives is shown in Scheme 1. First, electrochemical preparation of *N*-protected 2,3-dihydro-4-acetoxypiperidine **2**, followed by generation of π -allyl palladium **3** from **2** by Pd(OAc)₂/PPh₃ and then, successive umpolung of **3** mediated by Et₂Zn.³

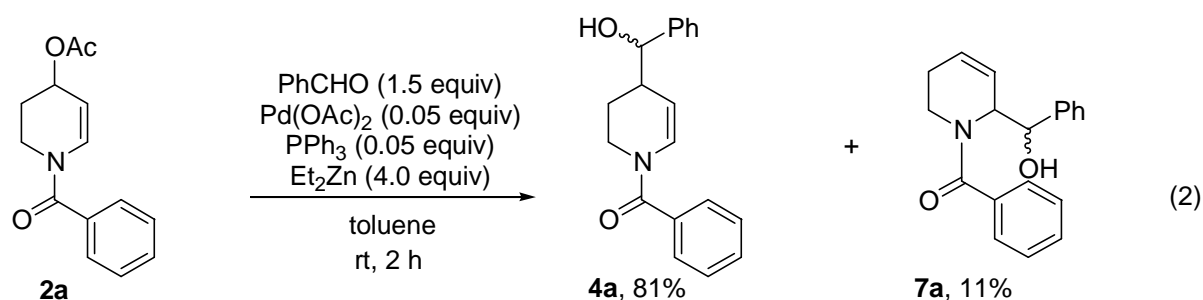
Scheme 1



Compounds **2** were prepared as follows (Eq. 1). Electrochemical oxidation of *N*-protected piperidines **1** afforded 2-methoxypiperidines **5**. Subsequent removal of methanol from **5**, followed by bromomethoxylation and dehydrobromination gave *N*-protected 2-methoxy-3,4-didehydropiperidines **6**,⁴ which were treated with AcOH to afford compounds **2** quantitatively.



With *N*-benzoyl-2,3-didehydro-4-acetoxypiperidine (**2a**)⁵ in hand, we first examined the reaction of **2a** with benzaldehyde using a catalytic amount of Pd(OAc)₂/PPh₃ in the presence of excess Et₂Zn in toluene (Eq. 2).⁶ The reaction proceeded smoothly within 2 h to afford 4-substituted piperidine **4a** as a major product in 81% and 2-substituted **7a** as a minor product in 11% yields.



In order to improve the regioselectivity, we screened a variety of *N*-protecting groups of **2** shown in Table 1 (Eq. 3). *p*-Chlorobenzoylated piperidine **2b** or *p*-trifluoromethylbenzoylated **2c** mainly afforded 4-substituted piperidine **4b** or **4c** along with some amount of 2-substituted **7b** or **7c**, respectively (entries 1 and 2). However the reaction of *p*-nitrobenzoylated one (**2d**) with benzaldehyde did not proceed at all (entry 3). On the other hand, compound **2e** protected with *p*-methoxybenzoyl group gave exclusively 4-substituted piperidine **4e** in excellent yield (entry 4), and **2f** protected with methoxycarbonyl group also gave 4-substituted **4f** in moderate yield (entry 5).

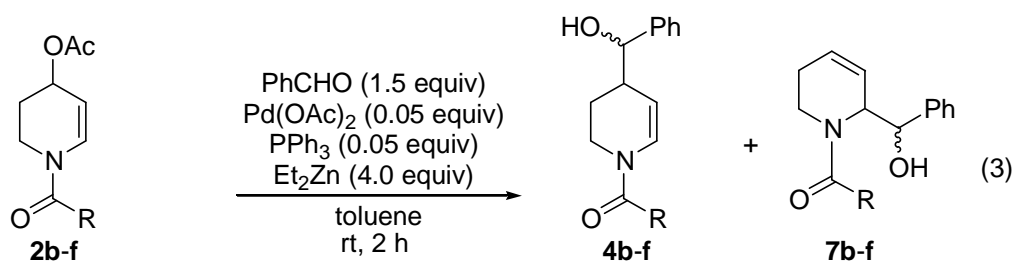
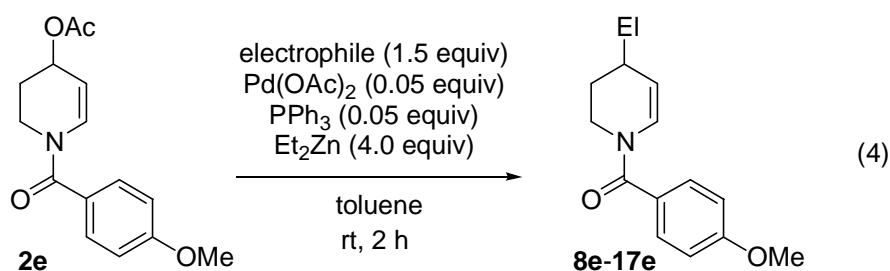


Table 1. Effect of *N*-protecting group on regioselectivity

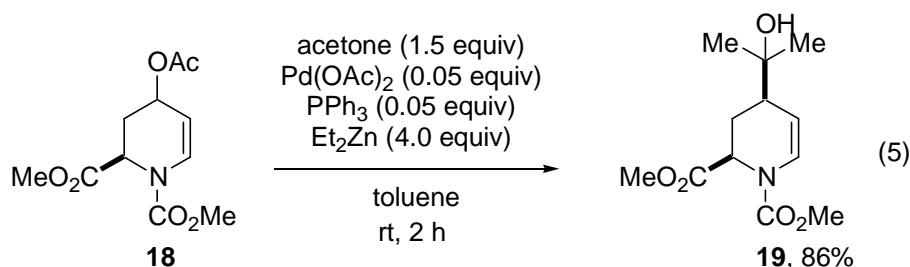
entry	4-acetate	R	product (yield: %)	
1	2b	<i>p</i> -ClC ₆ H ₄	4b (71)	7b (8)
2	2c	<i>p</i> -CF ₃ C ₆ H ₄	4c (66)	7c (13)
3	2d	<i>p</i> -NO ₂ C ₆ H ₄	4d (0)	7d (0)
4	2e	<i>p</i> -MeOC ₆ H ₄	4e (93)	7e (0)
5	2f	OMe	4f (54)	7f (0)

Next, the electrophilic substitution of **2e** with various electrophiles was examined (Eq. 4). These results are summarized in Table 2. Some aromatic (entries 1-3) and aliphatic aldehydes (entry 4) gave the corresponding coupling products **8e-11e** in good yields. Styrene oxide, which was transformed into phenylacetaldehyde under the reaction conditions, afforded **12e** in 80% yield (entry 5). Moreover, acyclic (entries 6-8) and cyclic ketones (entry 9) gave 4-substituted products **13e-16e** in good to high yields, while benzylideneaniline gave amine **17e** in high yield (entry 10).

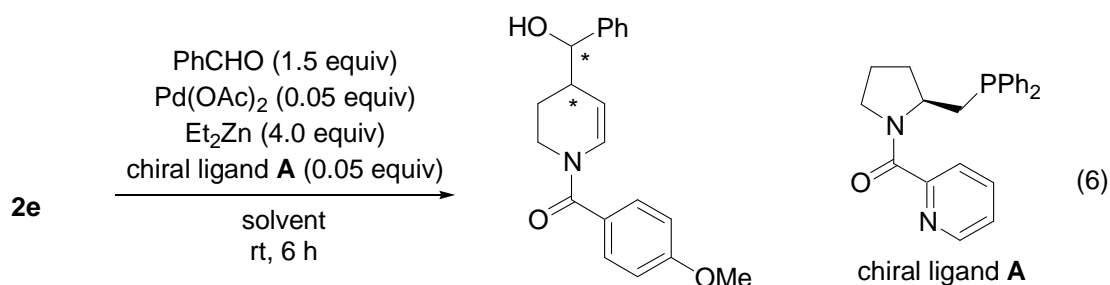
**Table 2.** Introduction of various electrophiles into **2e**

entry	electrophile	product		entry	electrophile	product	
			(yield: %)				(yield: %)
1	<i>p</i> -MeC ₆ H ₄ CHO		8e (70)	6		13e (86)	
2	<i>p</i> -ClC ₆ H ₄ CHO		9e (67)	7		14e (78)	
3	2-furyl-CHO		10e (70)	8		15e (65)	
4	<i>i</i> -Pr-CHO		11e (69)	9		16e (64)	
5	styrene oxide		12e (80)	10		17e (81)	

The reaction of pipercolinic acid derivative **18** with acetone proceeded regio- and stereo-selectively to afford *cis*-2,4-disubstituted product **19** in high yield (Eq. 5).⁷ The relative stereoconfiguration of **19** was deduced by NOE correlation.⁸



Chiral phosphine ligand **A**⁹ was used to introduce chirality in product **4e**.¹⁰ Use of toluene as a solvent gave diastereomer mixture of **4e** in low enantioselectivities, while CH₂Cl₂ led to moderate improvement in enantioselectivities of **4e** (Eq. 6).¹²



solvent	yield	diastereomer ratio and ee
toluene	70%	75 (11% ee) : 25 (20% ee)
CH ₂ Cl ₂	62%	80 (44% ee) : 20 (36% ee)

In summary, efficient regioselective introduction of various electrophiles into piperidine skeleton at the 4-position was achieved with a catalytic amount of Pd(OAc)₂/PPh₃ in the presence of excess Et₂Zn. In addition, enantioselective introduction of benzaldehyde into **2e** at the 4-position was accomplished by use of chiral phosphine ligand **A** with moderate enantioselectivity. Further improvement of diastereo- and enantio-selectivity is underway.

ACKNOWLEDGEMENTS

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REFERENCES AND NOTES

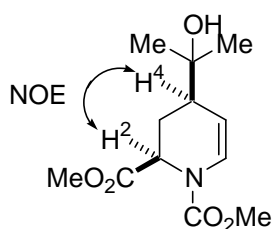
[†] Dedicated to Professor Ryoji Noyori on the occasion of his 70th birthday.

1. For recent examples, see: C. De Risi, G. Fanton, G. P. Pollini, C. Trapella, F. Valente, and V. Zanirato, *Tetrahedron: Asymmetry*, 2008, **19**, 131; G. S. Kauffman, P. S. Watson, and W. A. Nugent, *J. Org. Chem.*, 2006, **71**, 8975; L. F. Solares, I. Lavandera, V. Gotor-Fernández, R. Brieva, and V. Gotor, *Tetrahedron*, 2006, **66**, 3284; Y. Mi and E. J. Corey, *Tetrahedron Lett.*, 2006, **47**, 2515; K. Tanaka, T. Kobayashi, H. Mori, and S. Katsumura, *J. Org. Chem.*, 2004, **69**, 5906; I. T. Raheem, S. N. Goodman, and E. N. Jacobsen, *J. Am. Chem. Soc.*, 2004, **126**, 706.
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5. Characterization data of **2a**: Colorless oil. IR (neat): 3447, 2937, 1738, 1645, 1578 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 1.92-2.21 (m, 5H), 3.41-3.53 (m, 1H), 4.28 (br s, 1H), 5.00 (br s, 1H), 5.20-5.29 (m, 1H), 6.68 (br s, 1H), 7.29-7.57 (m, 5H). MS [HR-FAB(+)]: *m/z* calcd for C₁₄H₁₆NO₃ 246.1130 [M+H]⁺ found 246.1108.
6. A typical experimental procedure: A solution of piperidine derivative **2a** (0.3 mmol, 73.5 mg), Pd(OAc)₂ (0.015 mmol, 3.4 mg), PPh₃ (0.015 mmol, 3.4 mg), 1M Et₂Zn in hexane (1.2 mmol, 1.2 mL), and benzaldehyde (0.45 mmol, 48 mg) in toluene (2.0 mL) was stirred for 2 h under a nitrogen atmosphere. The resulting mixture was poured into saturated aqueous NH₄Cl and extracted with AcOEt (10 mL x 3). The combined organic layer was dried over MgSO₄ and concentrated in vacuo, the residue was chromatographed on silica gel (hexane/AcOEt = 3/1) to afford **4a** in 81% and **7a** in 11% yield as colorless oil, respectively. **4a**: IR (neat): 3450, 2920, 1655, 1490 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) δ 1.92-2.10 (m, 2H), 2.52-2.65 (m, 1H), 3.31-3.42 (m, 1H), 3.50-3.63 (m, 1H), 3.95-4.13 (m, 1H), 4.45-4.51 (m, 1H), 5.08-5.15 (m, 1H), 6.45-6.55 (m, 1H), 7.20-7.61 (m, 10H). MS [HR-FAB(+)]: *m/z* calcd for C₁₉H₂₀NO₂ 294.1494 [M+H]⁺ found 294.1493. **7a**: IR (neat): 3420,

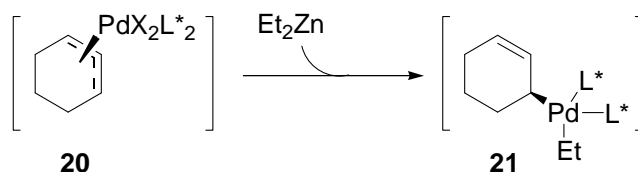
2931, 1716, 1645 cm^{-1} . ^1H NMR (300 MHz, CDCl_3) δ 1.43-1.73 (m, 1H), 2.16-2.27 (m, 1H), 3.13-3.25 (m, 1H), 3.25-3.47 (m, 2H), 4.39-4.53 (m, 1H), 4.81-4.92 (m, 2H), 5.82-5.88 (m, 1H), 7.20-7.61 (m, 10H).

7. Characterization data of **19**. Colorless oil. IR (neat): 3504, 2959, 1716, 1655, 1448 cm^{-1} . ^1H NMR (300 MHz, CDCl_3) δ 1.19 (s, 3H), 1.22 (s, 1.2H), 1.23 (s, 1.8H), 1.45 (br s, 1H), 1.72-1.84 (m, 1H), 2.08-2.14 (m, 1H), 2.39-2.47 (m, 1H), 3.75 (s, 3H), 3.76 (s, 1.2H), 3.80 (s, 1.8H), 4.82-4.85 (m, 0.4H), 4.90 (d, $J=8.5$ Hz, 0.6H), 4.97-5.00 (m, 1H), 6.87 (d, $J=8.5$ Hz, 0.6H), 7.00 (d, $J=8.5$ Hz, 0.4H). MS [HR-FAB(+)]: m/z calcd for $\text{C}_{12}\text{H}_{20}\text{NO}_5$ 258.1341 $[\text{M}+\text{H}]^+$ found 258.1339.

8. NOE correlation was observed between H^2 and H^4 .



9. K. Hiroi, Y. Suzuki, and I. Abe, *Tetrahedron: Asymmetry*, 1999, **10**, 1173.
10. It was proposed in ref 11 that a plausible intermediate in the asymmetric reaction of cyclohexenyl acetate with benzaldehyde might be η^1 -allylpalladium species **21** generated from η^3 -allylpalladium species **20** with Et_2Zn .



11. G. P. Howell, A. J. Minnaard, and B. L. Feringa, *Org. Biomol. Chem.*, 2006, **4**, 1278.
12. Characterization data of **4e** obtained in CH_2Cl_2 (The absolute stereoconfiguration is not determined). Colorless oil. $[\alpha]_D^{19} -9.1$ (c 1.07, CHCl_3). IR (neat): 3420, 2934, 1732, 1651 cm^{-1} . ^1H NMR (300 MHz, CDCl_3) δ 1.70 (br s, 1H), 1.99 (br s, 2H), 2.59-2.64 (m, 1H), 3.52-3.61 (m, 1H), 3.84 (s, 3H), 3.99-4.04 (m, 1H), 4.43-4.58 (m, 1H), 5.05-5.19 (br s, 1H), 6.60 (br s, 1H), 6.90 (d, $J=8.7$ Hz, 2H), 7.22-7.40 (m, 5H), 7.45 (d, $J=8.7$ Hz, 2H). MS [HR-FAB(+)]: m/z calcd for $\text{C}_{20}\text{H}_{22}\text{NO}_3$ 324.1600 $[\text{M}+\text{H}]^+$ found 324.1598. The diastereoselectivity and optical purity of **4e** were determined by chiral HPLC: Daicel Chiralcel OJ-H column (4.6 mm ϕ , 250 mm), n -hexane : i -PrOH = 3 : 1, wavelength: 254 nm, flow rate: 1.0 mL/min, retention time: Major diastereomer 12.9 min (rich), 22.9 min and minor diastereomer 27.5 min (rich), 38.5 min.

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