

1 **Prediction model for anti-malarial activities of hemozoin inhibitors using**  
2 **physicochemical properties**

3 **Running title:** *“Prediction model for anti-malarial activities of hemozoin inhibitors”*

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56 **Abstract**

57 The rapid spread of strains of malaria parasites resistant to several drugs has threatened global  
58 malaria control. Hence, the aim of this study was to predict the anti-malarial activity of  
59 chemical compounds possessing anti-hemozoin formation activity as a new means of anti-  
60 malarial drug discovery. After the initial *in vitro* anti-hemozoin formation high-throughput  
61 screening (HTS) of 9,600 compounds, a total of 224 hit compounds were identified as  
62 hemozoin inhibitors. These 224 compounds were tested for *in vitro* erythrocytic anti-malarial  
63 activity at 10  $\mu$ M using the chloroquine-mefloquine sensitive *Plasmodium falciparum* strain,  
64 3D7A. Two independent experiments were conducted. The physicochemical properties of the  
65 active compounds were extracted from ChemSpider and SciFinder databases. We analyzed the  
66 extracted data using Bayesian model averaging (BMA). Our findings revealed that lower  
67 numbers of S atoms, lower values of log D pH 3, 4 and 5, and higher values of ACD log D pH  
68 7.4 had a significant association with anti-malarial activity among compounds possessing anti-  
69 hemozoin formation activity. The BMA model revealed an accuracy of 91.23%. We report new  
70 prediction models containing the physicochemical properties that shed light on effective  
71 chemical groups for synthetic anti-malarial compounds and help *in silico* screening for novel  
72 anti-malarial drugs.

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## 79 **Introduction**

80 Despite the recent advances in anti-malarial drug discovery and development, malaria remains one  
81 of the most serious medical burdens worldwide, resulting 212 million new cases with 429,000  
82 deaths in 2015 (1). According to the different stages of the parasite life cycle, several therapies can  
83 affect its stages and exert their therapeutic roles. Quinine, amodiaquine, chloroquine, and  
84 mefloquine, which contain a quinoline scaffold, are fast-acting and highly effective blood  
85 schizontocidal therapies against malaria parasites, including *Plasmodium falciparum* (*PF*) (2).

86 Indeed, the resistance of malaria to chloroquine and other 4-aminoquinoline-based therapies, in  
87 addition to the antifolate combination sulfadoxine/pyrimethamine, has turned the light on  
88 artemisinin-based combinations to achieve higher response rates (3, 4). However, the rapidly  
89 spreading resistance of *PF* to artemisinin-based combinations has been reported causing a global  
90 challenge for malaria control (5, 6). Thus, it is important to discover new anti-malarial drugs,  
91 especially for malaria-endemic countries.

92 Recently, several new classes of anti-malarials have entered clinical studies, in patients with malaria,  
93 such as the fast-acting agents KAF156 (7), cipargamin (8), and artefenomel (9), whereas ferroquine  
94 remains the only long-acting novel anti-malarial in clinical development (10, 11). However, these  
95 drugs have not been approved yet and no vaccine to help in prevention, control, elimination, and  
96 eradication of malaria has been approved yet. Only one vaccine candidate, RTS, S/AS01, reached  
97 phase-III clinical trials with relatively low efficacy (12–14). Therefore, there is an urgent need for  
98 the discovery and development of novel anti-malarial chemotherapies for which there are no  
99 preexisting resistance mechanisms.

100 As of today, one of the most promising and ideal targets is interference with the parasite's heme  
101 detoxification pathway which is the target for some current anti-malarial drugs like quinine that is  
102 still efficacious against chloroquine-resistant *PF* (15–19). Recently, inhibition of the heme

103 detoxification pathway of the parasite has been highlighted as a target in several anti-malarial  
104 screening projects (20–22). This target is based on the inhibition of hemozoin which is a crystalline  
105 pigment produced by the malaria parasite as a result of the hemoglobin degradation process to  
106 protect it against the toxic heme produced as an end product of hemoglobin catabolism (23, 24).

107 Hemozoin formation is a protective physiochemical process that needs parasite protein (25–27)  
108 and/or food vacuole lipids or membranes (28, 29) for synthesis. Therefore, lipophilic detergents  
109 that mimic the intra-parasite condition like Nonidet P-40 and Tween 20 can be used as surrogate  
110 substances for high-throughput screening (HTS) of novel anti-malarials because they have the  
111 ability to promote the crystallization of heme (20, 30). This makes hemozoin inhibition suitable for  
112 research using HTS assays to build prediction models for novel anti-malarial drugs.

113 Recently, several studies used HTS and predicted models for  $\beta$ -hematin, synthetic hemozoin,  
114 inhibitors. Sandlin *et al* have screened 144,330 and produced 530 hits of which 171 were active  
115 against parasites and 73 had parasite  $IC_{50}$ s below 5  $\mu$ M and 25 below 1  $\mu$ M. (31). In addition, using  
116 physiochemical properties (22), we have recently developed an *in silico* model to predict drug-like  
117 compounds possessing anti-hemozoin activity.

118 As previously suggested, prediction models possess advantages for anti-malarial design because  
119 other approaches such as analog development based on existing agents or natural products mainly  
120 detect new anti-malarials by the chemical modifications of previously known compounds (32),  
121 however, new anti-malarial compounds can be discovered by the prediction equation based on a  
122 well-known metabolic target. Thus, the prediction models aid discovery of new chemical scaffolds.  
123 Moreover, specialized labware and expensive equipment are not required in these models, so  
124 millions of library compounds can be screened in-silico by using the prediction models. Also, the  
125 relationship between the compound's properties and anti-hemozoin activity is interpreted from the  
126 prediction models. Therefore, we continued the previous work by developing new prediction

127 models for novel anti-malarial activities of hemozoin inhibitors using the physiochemical properties  
128 of these small chemical compounds.

## 129 **Methods**

### 130 *Ethics statement*

131 Experiments which required human materials, red blood cell (RBC) and serum, were approved by  
132 the institutional ethical review board of Institute of Tropical Medicine (NEKKEN), Nagasaki  
133 University.

### 134 *Materials*

135 Chloroquine (Sigma-Aldrich Chemical Company, UK) was used to monitor the assay system,  
136 stored in 4°C freezer and mixed well before use. Dimethyl sulfoxide (DMSO; Wako Pure  
137 Chemicals Ltd., Osaka, Japan) was used as negative control. SYBR Green I (10,000 × stock conc.)  
138 was obtained from Lonza (Lonza, Rockland, ME, USA) and stored in -30 °C. Alamar Blue was  
139 purchased from Funakoshi (Funakoshi Co., Tokyo, Japan). Lysis buffer containing Tris (20 mM;  
140 pH 7.5), EDTA (10 mM), saponin (0.01%, wt/vol) and Triton X-100 (0.1%; vol/vol) was prepared  
141 in advance and kept at 4 °C. Human O<sup>+</sup> RBC and serum were obtained from the Japanese Red Cross  
142 Society (Reception Number: 28J0060). Compounds, which were identified positive from our  
143 previous study (22) were obtained from the Open Innovation Center for Drug Discovery (OCDD),  
144 University of Tokyo (Tokyo, Japan). All compounds received from the University of Tokyo at 10  
145 μM concentration, however, the stock solutions were prepared in DMSO at 2 mM concentration.  
146 The final DMSO concentration used in the experiment was ≤ 0.5% which had no inhibitory effect  
147 on parasite culture.

### 148 *Parasite culture*

149 For this study, chloroquine-mefloquine sensitive *PF* strain 3D7A was used. *PF* strain (3D7A) was  
150 cultured and maintained as previously described (33) with slight modifications (34). In brief,

151 parasites were maintained in RPMI 1640 (Thermo Fisher Scientific, MA, USA) medium  
152 supplemented with 5% AB<sup>+</sup> human serum, 0.25% Albumax I (Thermo Fisher Scientific), 12.5  
153 µg/ml gentamicin (Sigma-Aldrich) and human RBC (O<sup>+</sup>) at a hematocrit level of 2% and 0.2 – 2%  
154 parasitemia in culture flasks at 37°C under an atmosphere of 5% CO<sub>2</sub>, 5% O<sub>2</sub> and 90% N<sub>2</sub>.

#### 155 *Asexual anti-malarial assay*

#### 156 ***In vitro* assay using chloroquine-mefloquine sensitive *PF* strain, 3D7A**

157 The erythrocytic anti-malarial activity of the compounds was measured by growth inhibition (%)  
158 of the parasites in the presence of compounds using SYBR Green I assay. In brief, 50 µL of *PF*  
159 strains were seeded into 96 well clear bottom black plate (Thermo Fisher Scientific) with 0.75%  
160 parasitemia and 2% hematocrit followed by addition of 50 µl of the compound at 10 µM final  
161 concentration. Wells, with the absence of parasite, are considered as a positive control and parasites  
162 in DMSO (≥ 5%) without any treatment considered as negative control. Plates were then incubated  
163 for 48 hours at 37°C with 5% CO<sub>2</sub>, 5% O<sub>2</sub>, and 90% N<sub>2</sub> using a closed Jar. After 48 hours of  
164 incubation at standard culture condition, 100 µl of lysis buffer containing SYBR Green I (1× final  
165 concentration) were directly added to the wells. Plates were then placed on a shaker with the gentle  
166 mix to incubate for 1 hour at room temperature in the dark. The fluorescence of each well then  
167 measured using a multi-plate reader (ARVO1430; PerkinElmer, MA, USA) in a fluorescence  
168 detection mode (Ex = 485 nm, Em = 515 nm) for a 0.1-second exposure. Two independent  
169 experiments were performed under similar condition. Subsequently, an *in vitro* dose-response assay  
170 was performed.

#### 171 ***In vitro* anti-malarial dose-response assay**

172 Active compounds identified from the previous anti-malarial assay eventually goes for *in vitro*  
173 erythrocytic dose-response assay at ten different dilutions ranged between 0.5 nM and 10 µM to  
174 exclude any false positive compounds and measure the half maximal inhibitory concentrations



175 (IC<sub>50</sub>). Each experiment was conducted twice using 96 well clear bottom black plate and the  
176 fluorescence was measured as described above. Obtained data were analyzed to calculate the IC<sub>50</sub>  
177 value of each compound based on a typical sigmoid dose-response by using GraphPad Prism 6  
178 software.

#### 179 *Anti-hemozoin dose-response assay*

180 The IC<sub>50</sub> of potential hit compounds against hemozoin were further performed as previously  
181 described (22). Briefly, hemin solution (10 mM heme in DMSO and 100 mM acetate buffer) were  
182 incubated with detergent NP-40 and compounds (0 μM to 208 μM) for 250 minutes at 37°C,  
183 followed by addition of pyridine solution with 10 minutes shaking. Finally, the absorbance of each  
184 compound was measured at 405/705 nm. The IC<sub>50</sub> of each compound were calculated using  
185 GraphPad Prism version 6 software.

#### 186 *Physicochemical properties of 224 hemozoin inhibitors*

187 Physicochemical properties of chemical compounds were obtained from SciFinder Scholar  
188 Database (American Chemical Society, Washington, DC) or ChemSpider ([www.chemspider.com](http://www.chemspider.com))  
189 as previously described (22).

#### 190 *Statistical analysis*

191 We conducted blinded experiments without knowledge of compound' names or structures. The  
192 code was only opened after sending the anti-malarial and anti-hemozoin results to the Open  
193 Innovation Center for Drug Discovery, University of Tokyo. Moreover, the statistician who  
194 performed the analysis is not in the chemistry field and was not involved in the experiments.  
195 Variables of physical properties and compounds with many missing data including pKa<sub>2</sub>, vapor  
196 pressure (Torr), flash point (°C), enthalpy of vaporization (kJ/mol), and boiling point (°C) were  
197 excluded. Finally, 206 compounds were included in our analysis.

198 The outcome variable was the anti-malarial activity of the chemical compounds (coded as 1 = active,  
199 0 = inactive). The predictor variables were the physicochemical properties of our chemical  
200 compounds. We performed uni-variable logistic regression (LR) to examine the association  
201 between physicochemical properties and the aforementioned outcome. Subsequently, variables with  
202 p-values ( $P$ ) < 0.1 and/or with a prior significance, in our previous study (22) were included in the  
203 multi-variable analyses using Bayesian model averaging (BMA) to find the independent predictors  
204 of our outcome. We have multiplied index of refraction variable by 10 to have a smaller odds ratio  
205 (OR).

#### 206 *Development and validation of the prediction models*

207 We divided the original data randomly into training and testing sets with a ratio of 70:30, (nearly  
208 157 (including 17 positive compounds) and 67 compounds (including 5 positive compounds),  
209 respectively). The training data set was constructed to develop prediction models using the model.  
210 Results are shown as a posterior probability (PP) and Bayesian information criteria (BIC) of each  
211 model (35, 36) (Tables 2 and 3). The PP,  $P(B \neq 0)$ , is the probability that a predictor variable has an  
212 effect on the compound activity and a co-efficient is nonzero. Then, the best models were illustrated  
213 visually by depicting the variables included in them (Figure 1).

#### 214 *Comparison between the best models*

215 The sensitivity and specificity of the best models were calculated. The discriminatory powers of the  
216 best prediction models obtained from BMA were compared according to the area under the curve  
217 (AUC) from the receiver operating characteristic (ROC), and their accuracies (Figure 2). All  
218 analyses were performed using RStudio software version 1.0.44 (<https://www.rstudio.com/>).  $P$  <  
219 0.05 was considered statistically significant in all analyses. The data and R script can be obtained  
220 from MGK or NTH.

## 221 **Results**

### 222 *In vitro anti-malarial assay*

223 A total of 224 compounds having hemozoin inhibitory activity (22) were selected for *in vitro* anti-  
224 malarial assay. Among those, 30 compounds having  $\geq 45\%$  growth inhibitory activity at 10  $\mu\text{M}$   
225 concentration were further subjected to a dose-response assay to remove false positive compounds  
226 from initial screening (Figure 1), resulting only 22 compounds with a clear sigmoid dose-response  
227 curve to determine  $\text{IC}_{50}$  (Figure 2 and Table 1).

228 The interference of 22 compounds with SYBR Green 1 in the presence of malaria parasites was  
229 measured by culturing *Plasmodium falciparum* 3D7A for 48 h and then adding 22 compounds  
230 followed by addition of SYBR Green 1 with 1 h incubation in the dark. Subsequently, the  
231 fluorescence was measured by using a multiplate reader (ARVO1430; PerkinElmer, MA) in the  
232 fluorescence detection mode (excitation [Ex] at 485 nm and emission [Em] at 515 nm) for a 0.1-s  
233 exposure. The results showed all fluorescence variations were less than 10%, indicating that there  
234 was almost no interference of the compounds, artemisinin, and chloroquine with SYBR Green – 1  
235 fluorescence (Supplementary Figure). Furthermore, our results demonstrated that all 22 compounds  
236 possessed anti-hemozoin with  $\text{IC}_{50}$  ranging from 4.58 to 198.1  $\mu\text{M}$  (Table 1).

### 237 ***Development of prediction models for anti-malarial activities***

238 Number of C, and S atoms, H acceptors, H donor acceptor sum, log D pH 1, 2, 3, 4 and 5, mass  
239 solubility pH 1 and 3 and polarizability had  $P$  value  $< 0.1$  in the uni-variable LR, and log P, KOC  
240 pH 5.5 and 7.4, log D pH 5.5 and 7.4, density, refractive index, rule number of five violations,  
241 number of freely rotatable bonds, H bond donors, H, and O were significant from the previous study  
242 (22), while number of N atoms and surface tension were significant from the current analysis and  
243 the previous study, respectively (22) (Table 2).

244 The BMA analysis selected 37 models, from which the best 5 models had 5 variables, as the best  
245 predictors of the chemical compound activity, including the number of S atoms, log D pH 3, 4 and  
246 5, and ACD log D pH 7.4. (Figure 3, Tables 2 and 3). This reflected that the anti-malarial activity

247 was significantly associated with lower number of S atom, lower log D pH 3, 4, and 5, and higher  
248 ACD log D pH 7.4.

### 249 ***Validation of the prediction model***

250 The BMA model, with a cut-off of 0.09, revealed a sensitivity, specificity, and maximum accuracy  
251 of 60%, 69.23%, and 91.23%, respectively.

### 252 **Discussion**

253 Our BMA prediction model identified several physicochemical properties from which we can  
254 design to develop novel drugs. We found that a lower number of S atoms and lower log D value at  
255 pH 3, 4 and 5 and a higher ACD log D value at pH 7.4 could serve as good predictors while  
256 developing novel anti-malarial drugs.

257 Ignatushchenko *et al.* have found that the replacement of the ring oxygen by sulfur to produce 4, 5-  
258 dihydroxythioxanthone, results in a decreased inhibitory activity on hemozoin formation and  
259 parasite growth when compared to the corresponding xanthone homolog (37).

260 A low log D, a low lipophilicity through high inhibitory activities against hemo crystallization, at  
261 pH 3, 4 and 5 were significantly associated with a better anti-malarial activity. It is well known that  
262 the anti-malarial drugs must enter to the parasite food vacuole where the pH is around 4 – 5.5.  
263 Noteworthy, log P is an indicator of lipophilicity and is a component of the Lipinski's rule of five,  
264 a rule of thumb, to predict the drug-likeness. However, log D is a better indicator at specific pH  
265 environment. Because the hemozoin formation occurs inside the acidic food vacuole, such  
266 compounds are expected to decrease its lipophilicity after the uptake into the parasite food vacuole  
267 and to be accumulated within it.

268 Log P is widely used, however, it fails to take into account the variation in the lipophilicity of a  
269 drug regarding the ionic states present at key biological pH values. Given that the majority of  
270 commercial pharmaceuticals contain an ionizable moiety, Bhal *et al.* that log D is a better descriptor  
271 for lipophilicity in the context of the rule of five. It gives more physiologically relevant results,

272 subsequently declining the potential false-negatives number incorrectly eliminated in screening.  
273 They also showed that the adapted rule of five using log D instead of log P gives a notable  
274 improvement in the pass rate for compounds that have the desired lipophilicity at a relevant  
275 physiological pH (38).

276 In 11 p-aminopyridine anti-malarials a quadratic association between log the lipid accumulation  
277 ratio (LAR), anti-log log D 7.4, and log resistance index was confirmed while log (LAR/vacuolar  
278 accumulation ratio (VAR)) versus log resistance index for 12 was linear. Both might be able to  
279 predict the utility of structural adjustments (39).

280 This confirms for chloroquine-sensitive parasites, that while charged drug concentration in food  
281 vacuole increases as vacuolar pH decreases, the concentration of uncharged drug (base) in vacuolar  
282 membrane and other lipid sites in the digestive vacuole has a constant value for each individual  
283 drug, because the proportion of drug partitioned into lipid decreases as log D (the log of [drug]lipid  
284 / [drug]water at equilibrium) decreases with a reduced pH.

285 Log D can predict gastrointestinal absorption and lipophilic properties because it is a pH-dependent  
286 function. Calculated values demonstrated the high gastrointestinal track absorption (pH 3–7) and  
287 lipophilic properties. It was observed a good correlation between the calculated distribution  
288 coefficients at pH 7.4 and pH 5.2 (log D 5.2 and 7.4), and the found inhibition percentages for the  
289 tested compounds (40).

290 In addition, the anti-malarial activity increases with increasing the ACD log D value at pH 7.4 is  
291 most probably because this favors the entry of the compound into the RBC, parasite and ultimately  
292 the digestive vacuole. On the other hand, a low ACD log D at a low pH (3 – 5) disfavors the egress  
293 of the compound out of the digestive vacuole. The consequence would be a greater accumulation  
294 in the digestive vacuole and hence higher concentrations at the site of action. Indeed, anti-log {ACD  
295 log D (7.4) – ACD log D (5)} is an experimental measure of the vacuolar pH trapping. It would  
296 make sense that this would only apply in the case of hemozoin inhibitors because these are the

297 compounds that act in the digestive vacuole. Since researchers prefer to use {ACD log D (7.4) -  
298 ACD log D (5)} for each compound. This might weaken the correlation and undermine this  
299 argument. Hence, we built a new model with that variable, {ACD log D (7.4) - ACD log D (5)} to  
300 investigate its effect, however, we found it insignificant.

301 Noteworthy, we have overcome our previous limitation in which we could not find any heavy atoms  
302 playing an important role in the refractive index which investigate the characteristics of the material  
303 (22). The presence and even the quantity of some heavy atoms and functional groups with a high  
304 refractive index, as sulfur (41) have an important role in increasing the molar refraction. It is  
305 noteworthy that out of the 224 included compounds, there were 87 compounds have at least 1 S  
306 atom with a mean of 1 atom and 131 compounds have no S atoms. Moreover, among negative and  
307 positive compounds there were 83 compounds and 4 compounds have at least one atom,  
308 respectively. Additionally, among our positive anti-malarial compounds, we have four compounds  
309 with quinolines which have no S atoms, hence this suggests a negative association between the S  
310 atoms and the anti-malarial activity. However, our results did not show an association between the  
311 refractive index and high anti-malarial activities, suggesting the refractive index is related only to  
312 anti-hemozoin but not in the anti-malarial activity. Although our models have not been validated in  
313 external samples, it can be the first clue for understanding the mechanism of action of anti-malarials  
314 (22). Moreover, future studies need to validate our results with a larger number of compounds,  
315 especially the positive ones.

## 316 **Conclusion**

317 Using physicochemical properties of the compounds, we reported new prediction models that could  
318 predict anti-malarial activity for chemical compounds possessing anti-hemozoin Our findings  
319 revealed that lower number of S atoms, lower values of log D pH 3, 4 and 5, and higher value of  
320 ACD log D pH 7.4 were independent predictors of a higher anti-malarial activity. This information  
321 can be used for *in silico* screening and modifying chemical structures to develop anti-malarials.

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## 328 **Conflict of interest**

329 None.

## 330 **Author contributions**

331 FM, AAT, MF did the experiments under the supervision of ShuM, OK, NTH, and KH. FM, MGK,  
332 TVD, DVT, SaM, YT, AMA, HTNG, TND, LKH, and MTE extracted the data from ChemSpider  
333 and SciFinder. MGK and NLV performed the statistical analysis under the supervision of NTH and  
334 KH. FM, MGK, TVD, AMA, OK, TJE, and NTH interpreted the analysis wrote the manuscript. All  
335 authors revised and approved the final version.

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#### 438 **Figure legends**

439 **Figure 1. The workflow of this study.** First, 224 compounds that showed anti-hemozoin activity in  
440 the previous study (22) were underwent in vitro anti-malarial assay at 10  $\mu$ M. Second, 30  
441 compounds having  $\geq 45\%$  parasite inhibitory efficacy have underwent in vitro dose-response assay  
442 between 0.5 nM and 10  $\mu$ M. Then, 22 compounds having  $IC_{50} \leq 10 \mu$ M were identified.  
443 Subsequently, the main structures and physicochemical properties of these compounds searched by

444 *SciFinder. Finally, the prediction models were generated by the traditional approach versus the*  
445 *Bayesian approach using their physical and chemical properties.*

446 ***Figure 2. The initial in vitro anti-malarial screening of 224 compounds using chloroquine-***  
447 ***mefloquine sensitive P. falciparum 3D7A strain. The circle dots represent % of parasite***  
448 ***inhibition of 224 compounds including 30 compounds having parasite inhibition  $\geq 45\%$ .***

449 ***Figure 3. Models selected by Bayesian model averaging. Blue and red colors represent positive***  
450 ***and negative variable estimates, respectively while uncolored variables were not included in the***  
451 ***model. On the x-axis, models were listed in the order of the decline in the posterior probability.***

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469 **Table 1. Anti-malarial s and anti-hemozoin activitie of 22 positive compounds.**

| Serial no. | Compound ID | <i>P. falciparum</i> (3D7A) | Anti-hemozoin         |
|------------|-------------|-----------------------------|-----------------------|
|            |             | IC <sub>50</sub> (μM)       | IC <sub>50</sub> (μM) |
| 1          | UTDIF3A11   | 0.66 ± 0.10                 | 78.1                  |
| 2          | UTDIF4A14   | 8.15 ± 2.60                 | 34.67                 |
| 3          | UTDIF5A15   | 9.00 ± 1.40                 | 14.01                 |
| 4          | UTDIF6C12   | 8                           | 4.58                  |
| 5          | UTDIF7D16   | 10.50 ± 2.10                | 87.76                 |
| 6          | UTDIF8E4    | 10                          | 38.54                 |
| 7          | UTDIF9E10   | 1.54 ± 0.07                 | 53.44                 |
| 8          | UTDIF10E14  | 10.00 ± 1.40                | 41.18                 |
| 9          | UTDIF11F12  | 9                           | 36.16                 |
| 10         | UTDIF12F15  | 8.95 ± 1.30                 | 103.5                 |
| 11         | UTDIF13G5   | 9.26 ± 1.80                 | 30.69                 |
| 12         | UTDIF14I6   | 7.00 ± 1.40                 | 43.98                 |
| 13         | UTDIF15I15  | 1.01 ± 0.50                 | 25.96                 |
| 14         | UTDIF16J16  | 9.28 ± 2.40                 | 28.5                  |
| 15         | UTDIF17K7   | 4.80 ± 1.70                 | 198.1                 |
| 16         | UTDIF18L5   | 3.06 ± 1.30                 | 110                   |
| 17         | UTDIF19L16  | 6.80 ± 4.40                 | 29.04                 |
| 18         | UTDIF20M7   | 8.00 ± 2.80                 | 156                   |
| 19         | UTDIF21O9   | 0.56 ± 0.27                 | 18.3                  |
| 20         | UTDIF22P6   | 0.06                        | 42.98                 |
| 21         | UTDIF23P14  | 9.50 ± 0.70                 | 24.72                 |
| 22         | UTDIF24G15  | 9                           | 160                   |

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472 Table 2. Uni-variable logistic regression (LR) statistics of the physicochemical properties of the compounds.

| Predictors           | Uni-variable analysis     |              |
|----------------------|---------------------------|--------------|
|                      | OR (95% CI)               | P            |
| Number of C atoms    | <b>1.11 (0.98 – 1.25)</b> | <b>0.092</b> |
| Number of H atoms*   | 1.07 (0.98 - 1.17)        | 0.111        |
| Number of N atoms    | <b>1.5 (1.15 - 1.95)</b>  | <b>0.003</b> |
| Number of O atoms*   | 1.02 (0.77 - 1.35)        | 0.912        |
| Number of S atoms    | <b>0.45 (0.19 - 1.06)</b> | <b>0.067</b> |
| Number of F atoms    | 0 (0 - Inf)               | 0.992        |
| Number of Cl atoms   | 1.45 (0.67 - 3.16)        | 0.348        |
| Number of Br atoms   | 0 (0 - Inf)               | 0.991        |
| Average mass atoms   | 1.01 (0.997 - 1.01)       | 0.214        |
| Monoisotopic         | 1.01 (0.998 - 1.01)       | 0.192        |
| Density*             | 0.89 (0.07 - 11.81)       | 0.929        |
| Rotatable bonds*     | 1.03 (0.85 - 1.26)        | 0.753        |
| H acceptors          | <b>1.38 (1.08 - 1.77)</b> | <b>0.01</b>  |
| H Donors*            | 1.11 (0.76 - 1.61)        | 0.605        |
| H donor acceptor sum | <b>1.25 (1.03 - 1.52)</b> | <b>0.022</b> |
| KOC pH 1             | 1 (0.9999 - 1.0001)       | 0.647        |
| KOC pH 2             | 1 (0.9999 - 1.0001)       | 0.568        |
| KOC pH 3             | 1 (0.9998 - 1.0001)       | 0.499        |

|                      |                                      |                     |
|----------------------|--------------------------------------|---------------------|
| KOC pH 4             | 1 (0.9998 - 1.0001)                  | 0.457               |
| KOC pH 5             | 1 (0.9998 - 1.0001)                  | 0.469               |
| KOC pH 6             | 1 (0.9999 - 1.0001)                  | 0.562               |
| KOC pH 7             | 1 (0.9999 - 1.0001)                  | 0.538               |
| KOC pH 8             | 1 (0.9999 - 1.0001)                  | 0.65                |
| KOC pH 9             | 1 (0.9999 - 1.0001)                  | 0.603               |
| KOC pH 10            | 1 (0.9998 - 1.0001)                  | 0.526               |
| LogD pH 1*           | <b><u>0.73 (0.58 - 0.92)</u></b>     | <b><u>0.007</u></b> |
| LogD pH 2            | <b><u>0.74 (0.59 - 0.93)</u></b>     | <b><u>0.009</u></b> |
| LogD pH 3            | <b><u>0.73 (0.58 - 0.92)</u></b>     | <b><u>0.008</u></b> |
| LogD pH 4            | <b><u>0.72 (0.56 - 0.92)</u></b>     | <b><u>0.008</u></b> |
| LogD pH 5            | <b><u>0.76 (0.59 - 0.97)</u></b>     | <b><u>0.027</u></b> |
| LogD pH 6            | 0.85 (0.66 - 1.09)                   | 0.199               |
| LogD pH 7            | 0.94 (0.74 - 1.2)                    | 0.643               |
| LogD pH 8            | 0.9995 (0.79 - 1.27)                 | 0.997               |
| LogD pH 9            | 1.04 (0.82 - 1.32)                   | 0.756               |
| LogD pH 10           | 1.09 (0.86 - 1.38)                   | 0.481               |
| Log P*               | 0.85 (0.62 - 1.17)                   | 0.319               |
| Mass intrinsic       | 0.97 (0.79 - 1.19)                   | 0.762               |
| Mass solubility pH 1 | <b><u>1.002 (0.9999 - 1.003)</u></b> | <b><u>0.067</u></b> |
| Mass solubility pH 2 | 1.001 (0.9997 - 1.003)               | 0.106               |

|                                       |                                      |                     |
|---------------------------------------|--------------------------------------|---------------------|
| <b>Mass solubility pH 3</b>           | <b><u>1.002 (0.9998 - 1.003)</u></b> | <b><u>0.078</u></b> |
| <b>Mass solubility pH 4</b>           | 1.001 (0.9991 - 1.004)               | 0.24                |
| <b>Mass solubility pH 5</b>           | 1.002 (0.9993 - 1.005)               | 0.136               |
| <b>Mass solubility pH 6</b>           | 1.001 (0.998 - 1.004)                | 0.535               |
| <b>Mass solubility pH 7</b>           | 0.997 (0.99 - 1.01)                  | 0.616               |
| <b>Mass solubility pH 8</b>           | 0.97 (0.89 - 1.06)                   | 0.525               |
| <b>Mass solubility pH 9</b>           | 0.84 (0.57 - 1.24)                   | 0.376               |
| <b>Mass solubility pH 10</b>          | 0.97 (0.91 - 1.03)                   | 0.312               |
| <b>Mass solubility</b>                | 0.47 (0.076 - 2.86)                  | 0.409               |
| <b>Molar volume</b>                   | 1.01 (0.997 - 1.01)                  | 0.218               |
| <b>Molecular weight</b>               | 1 (0.997 - 1.01)                     | 0.283               |
| <b>pKa1</b>                           | 1.05 (0.93 - 1.17)                   | 0.444               |
| <b>Polar surface</b>                  | 1 (0.99 - 1.02)                      | 0.733               |
| <b>Total score</b>                    | 1.22 (0.82 - 1.82)                   | 0.334               |
| <b>Polarizability</b>                 | <b><u>1.07 (0.996 - 1.14)</u></b>    | <b><u>0.066</u></b> |
| <b>Surface tension</b>                | 1.02 (0.99 - 1.05)                   | 0.256               |
| <b>Refractive index<sup>#</sup></b>   | 1.36 (0.72 - 2.57)                   | 0.348               |
| <b>Rule number of five violations</b> | 2 (0.67 - 5.9)                       | 0.212               |
| <b>ACD logD pH 5.5*</b>               | 0.87 (0.68 - 1.11)                   | 0.256               |
| <b>ACD logD pH 7.4*</b>               | 1.07 (0.83 - 1.36)                   | 0.617               |



|                        |                     |       |
|------------------------|---------------------|-------|
| <b>ACD BCF pH 5.5</b>  | 1 (1 - 1)           | 0.771 |
| <b>ACD BCF pH 7.4</b>  | 1 (0.9998 - 1.0001) | 0.737 |
| <b>ACD KOC pH 5.5*</b> | 1 (1 - 1)           | 0.77  |
| <b>ACD KOC pH 7.4*</b> | 1 (0.9999 - 1.0001) | 0.766 |

473 Abbreviations: OR = odds ratio; CI = confidence interval;

474 \*Variables with prior significance in our previous study (22). Significant predictors are in bold and underscored.

475 #Refractive index was multiplied by 10 to get a smaller OR.

476

477 **Table 3. The best five parsimonious models selected by BMA.**

| Model No. | Predictor         | Coefficient of predictor | BIC of model | PP of model |
|-----------|-------------------|--------------------------|--------------|-------------|
| 1         | Intercept         | -2.21                    | -648.24      | 0.17        |
|           | Number of S atoms | -1.26                    |              |             |
|           | Log D pH 4        | -0.86                    |              |             |
|           | ACD Log D pH 7.4  | 0.78                     |              |             |
| 2         | Intercept         | -2.76                    | -647.46      | 0.11        |
|           | Log D pH 4        | -0.84                    |              |             |
|           | ACD Log D pH 7.4  | 0.81                     |              |             |
| 3         | Intercept         | -2.22                    | -646.94      | 0.09        |
|           | Number of S atoms | -1.3                     |              |             |
|           | Log D pH 3        | -0.78                    |              |             |
|           | ACD Log D pH 7.4  | 0.68                     |              |             |
| 4         | Intercept         | -2.05                    | -646.02      | 0.06        |
|           | Number of S atoms | -1.22                    |              |             |
|           | Log D pH 5        | -0.92                    |              |             |
|           | ACD Log D pH 7.4  | 0.85                     |              |             |
| 5         | Intercept         | -2.57                    | -645.61      | 0.05        |
|           | Log D pH 5        | -0.92                    |              |             |
|           | ACD Log D pH 7.4  | 0.88                     |              |             |

478 Abbreviations: No. = number; BIC = Bayesian information criterion; PP = posterior probability. The cumulative PP = 0.468.

Figure 1

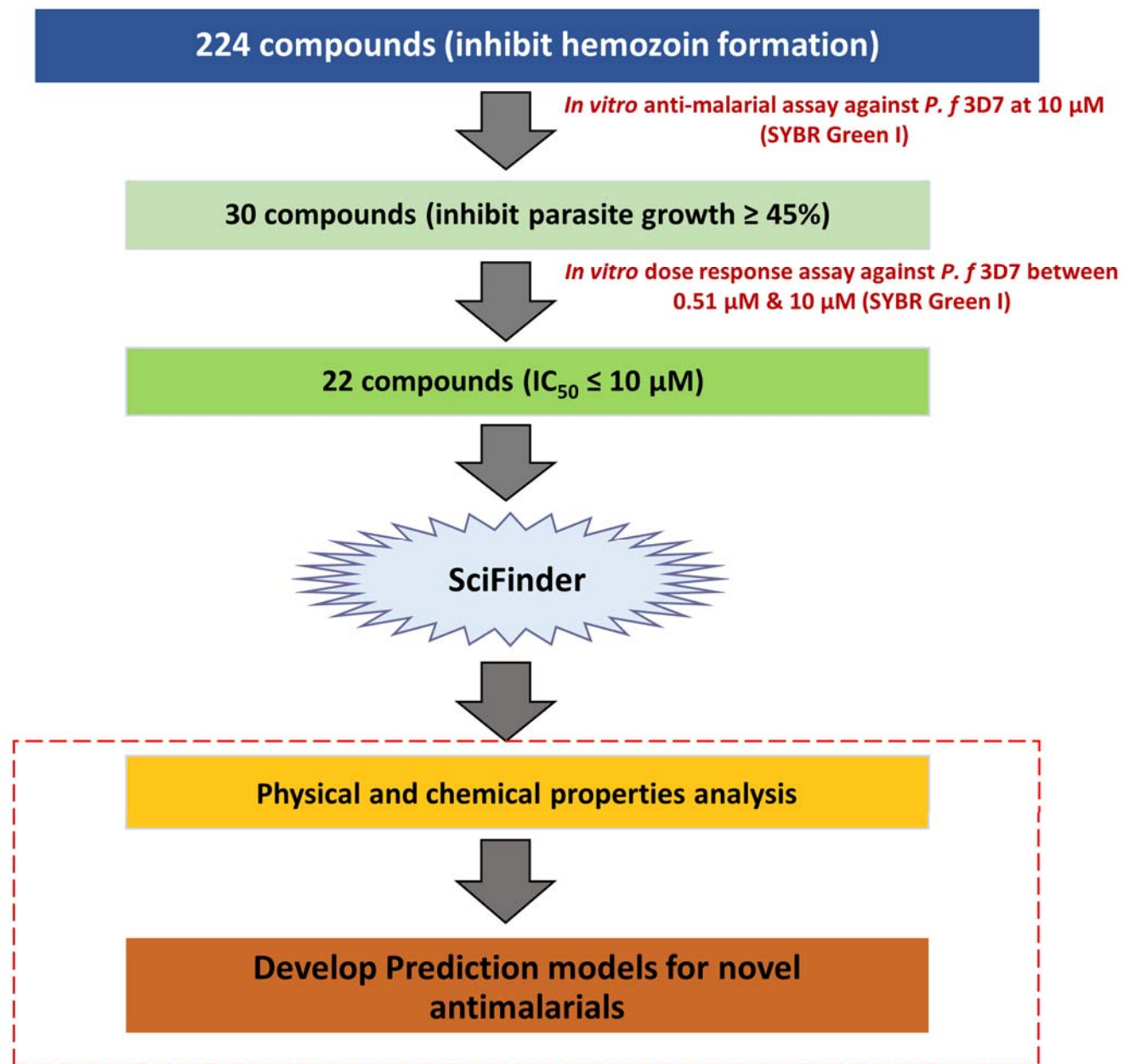
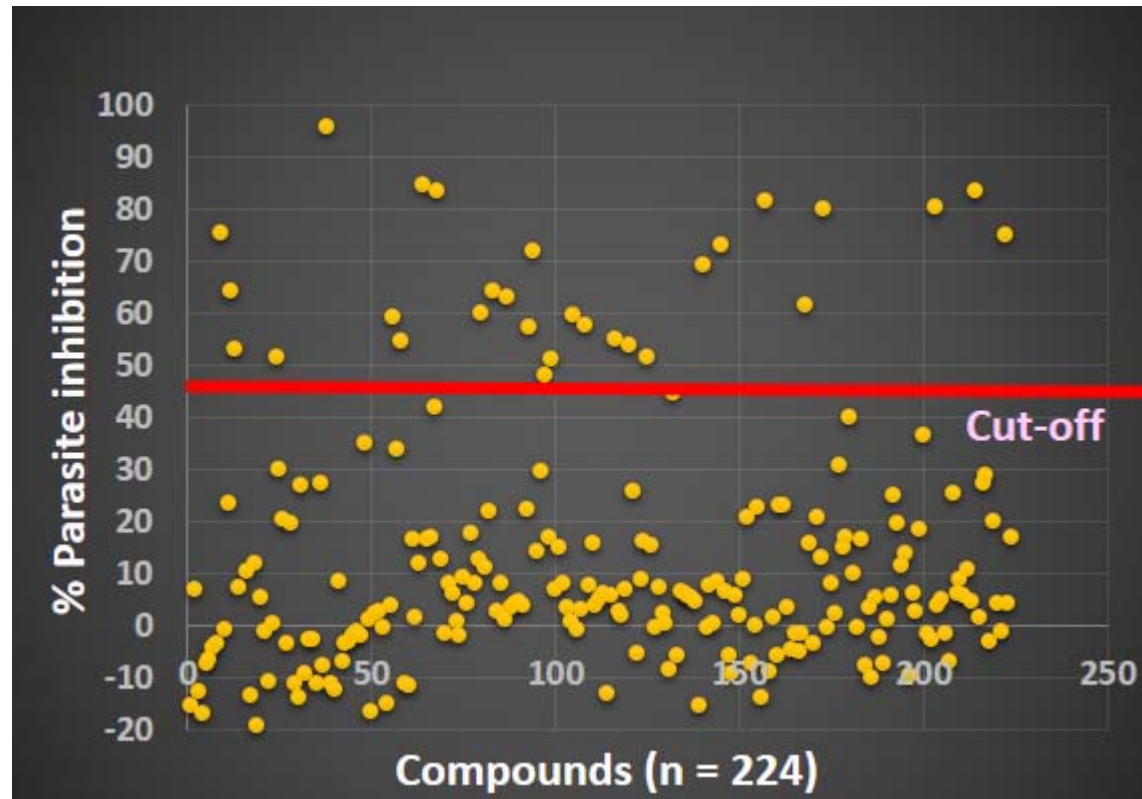


Figure 2

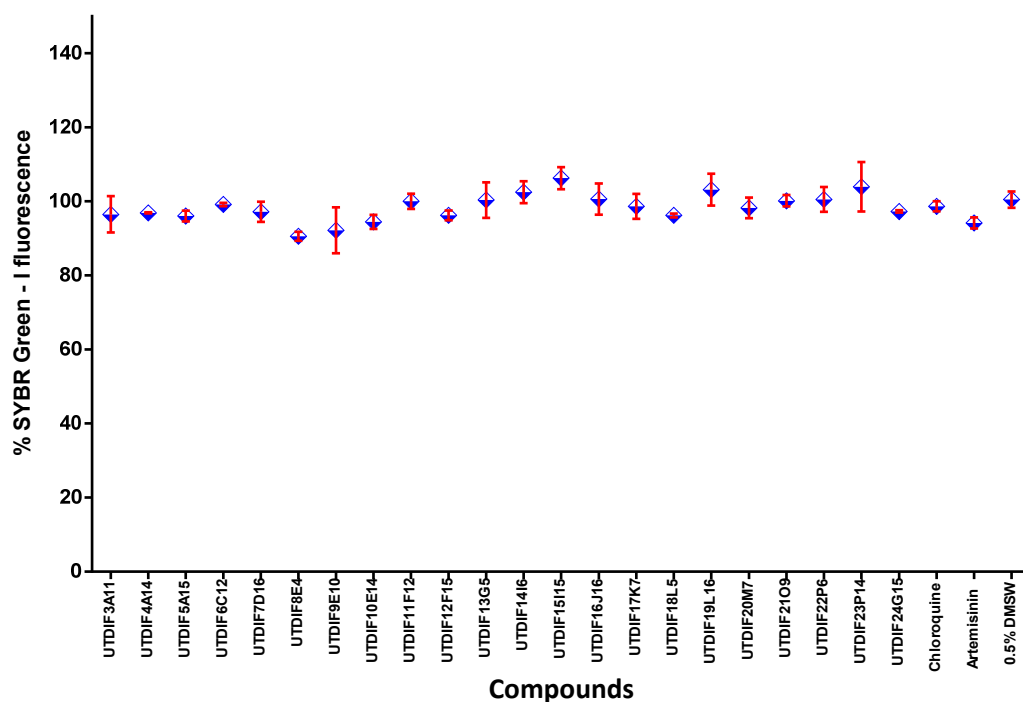




## Supplementary Data

**Method:** *Plasmodium falciparum* sensitive strain, 3D7A (3% parasitemia, 2% haematocrit) were seeded in 96-well black clear bottom plate. Subsequently, Chloroquine (10  $\mu$ M), Artemisinin (10  $\mu$ M), 22 compounds (10  $\mu$ M) and DMSO (0.5%) were added onto the parasites (3D7A) in 96-well black clear bottom plate. Afterwards, lysis buffer containing SYBR Green – I was added, incubate for 1 hour in dark and measure the fluorescence using Perkin-Elmer plate reader.

**Results:** The % of fluorescence obtained from compounds & controls (positive – Artemisinin & Chloroquine, negative – 0.5% DMSO) were measured to check the interference of compounds with SYBR Green – I fluorescence. The negative control (0.5% DMSO) exhibited 100% fluorescence. Other 22 compounds and positive controls showed >90% fluorescence which indicates that there was almost no interference of the compounds with SYBR Green – I fluorescence.



**Fig A:** The % of fluorescence of 22 compounds, Artemisinin & Chloroquine (positive control) and 0.5% DMSO (negative control) have shown. The % of fluorescence of all compounds are compared with the % of fluorescence of 0.5% DMSO (negative control).