E-Selective Ring-closing Metathesis in α-Helical Stapled Peptides Using Carbocyclic α,α-Disubstituted α-Amino Acids

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Supporting Information Placeholder



ABSTRACT: We present an *E*-selective ring-closing metathesis reaction in α -helical stapled peptides at *i* and *i* + 4 positions. The use of two chiral carbocyclic α, α -disubstituted α -amino acids, (1*S*,3*S*)-Ac₅c^{3OAII} and (1*R*,3*S*)-Ac₅c^{3OAII} provide a high *E*-selectivity of up to a 59:1 *E/Z* ratio, while mixtures with a *E/Z* of 2.1–0.5:1 were produced with standard acyclic (*S*)-(4-pentenyl)alanine amino acids. A stapled octapeptide composed of (1*S*,3*S*)-Ac₅c^{3OAII} amino acids showed a right-handed α -helical crystal structure.

The function of a protein strictly depends on its higher-order structure formed by protein folding. To achieve a similar functionality using short oligopeptides, it is important to constrain their secondary structures to a specific state. This issue can be resolved by the concept of foldamers.¹ Helical foldamers, including peptides,² oligoureas,³ and polymers,⁴ are representative foldamers that can be used as organocatalysts, chiral ligands, drug delivery tools, etc. The hydrocarbon stapling of a peptide is a powerful and convenient approach to stabilize the secondary structure of helical peptide foldamers and to enhance their functionality.⁵ The first example of helical hydrocarbonstapled peptide was reported by Blackwell and Grubbs using Oallyl L-homoserine-containing peptide as a cyclization precursor (Figure 1A).⁶ The ring-closing metathesis (RCM) reaction of this peptide using the first-generation Grubbs catalyst gave the desired 3₁₀-helical stapled peptides in high yields as an inseparable E/Z mixture with a ca. 5:1 ratio. In 2000, Verdine and co-workers reported all-hydrocarbon stapling with (S)-(4-pentenyl)alanines [Ala(4-Pte)] at i and i + 4, as well as i + 7 positions in α -helical peptides (Figure 1B).⁷ The all-hydrocarbon stapling became one of the most commonly used strategies to induce helicity, although their *E*/*Z*-selectivities as well as crystallographic structures have not been elucidated. A joint group of Toniolo, Grubbs, and O'Leary reported the first example of *E*-selective hydrocarbon-stapling formation with a E/Z of over 20:1 at *i* and *i* + 3 positions in 3₁₀-helical peptide foldamers (Figure 1C).⁸ They used *O*-allyl L-serine in a peptide sequence of α -aminoisobutyric acid (Aib), which preferred a 3₁₀-helical secondary structure.⁹ Grubbs, O'Leary et al. reported *Z*-selective all-hydrocarbon stapling in α -helical peptides using a *Z*-selective ruthenium catalyst (*E*:*Z* = <1:9).¹⁰ However, there are no reports on *E*-selective hydrocarbon-stapling at *i* and *i* + 4 positions of peptides, while development of *E*-selective macrocyclization by RCM has attracted increased attention in the last



Figure 1. Helical stapled peptides obtained by the RCM reaction (^{*a*}crystallographic structure of peptides, ^{*b*}conversion, ^{*c*}the *E*/*Z*-ratio was not determined)

decade.¹¹ A successive RCM reaction at *i* and *i* + 4 positions of peptides followed by ethenolysis of the undesired *Z*-configured staples reported by Grubbs et al. might be an alternative route to obtain *E*-enriched staples with a *E*/*Z* of up to 24:1.¹² However, the initial *E*/*Z*-selectivities of the RCM products were moderate between 2.4:1 to 4.9:1. Since α -helix is the most abundant secondary structure, α -helix-inducing *i*,*i* + 4-stapled peptides are frequently used for the peptide-based drug discovery.¹³ Furthermore, recent studies imply that *E*/*Z*-configuration of stapled peptides is a key to their biological activity.¹⁴ Accordingly, a direct approach to access the *E*-selective hydrocarbon-stapled peptides at *i*,*i* + 4 positions has been demanded.¹⁵

A possible reason of poor *E*-selectivities of hydrocarbon-stapling at *i* and i + 4 positions is a flexible nature of the tethered side chains of L-homoserine and (S)-Ala(4-Pte), which easily form both E- and Z-hydrocarbon stapling. If the conformation of tethered side chains of the stapling precursor is properly constrained, selective hydrocarbon-stapling of thermodynamically more stable *E*-isomer at *i* and i + 4 positions in α -helical peptides is possible. Accordingly, carbocyclic α, α -disubstituted α amino acids (dAAs)^{16,17} are suitable for this purpose because of a rigid carbocyclic ring and existence of the second stereogenic center on the ring. Furthermore, carbocyclic dAAs enhance the efficacy of peptide coupling reaction,^{9b,18} while side-chain stereogenic centers control a helical screw direction, as well as a secondary structure of their peptides.¹⁹ We have recently reported the synthesis of L-arginine-enriched stapled peptides containing 3-allyloxy-1-aminocyclopentanecarboxylic acids (Ac_5c^{3OAll}) possessing a high cell permeability, as well as a high stability against peptidases.²⁰ However, a detailed structural analysis of peptides has not been performed, particularly for their *E*/*Z*-selectivity and crystallographic structures. This work is aimed at the synthesis of stapled L-leucine-based Ac₅c^{3OAll}containing peptides and determination of the impact of Ac₅c^{3OAll} stereochemistry. A peptide containing Verdine's Ala(4-Pte) was also synthesized as a control. Herein, we report E-selective RCM in α -helical peptides stapled at *i* and *i* + 4 positions and analysis of their structure by X-ray diffraction (Figure 1D).

We synthesized four Ac_5c^{3OAII} (Figure S1) from (1*S*,3*S*)- and (1*R*,3*S*)-1-amino-3-hydroxycyclopentanecarboxylic acids.^{21,22} These Ac_5c^{3OAII} were successfully introduced to L-leucinebased peptides to provide **1a** octapeptides as RCM precursors.²² Further, RCM between tethered allyloxy groups of compound **1a** was performed by Grubbs catalyst to produce stapled peptides **2a**. After optimization of the reaction conditions (see Table S1),²² the reaction of **1a** using 20 mol % of the second-generation Grubbs catalyst in toluene completed within 30 min gave **2a** with an excellent *E/Z* of 24:1 and 96 % isolated yield. The amount of catalyst could be reduced to 10 mol % preserving a high *E*-selectivity of 21:1.

Under the optimized conditions, however, other peptides **1b**-**1d** possessing different stereochemistry on Ac_5c^{3OAII} provided moderate to low *E*-selectivities (E/Z = 1.9-13:1, Scheme 1). Moreover, considerable amounts of dimer byproducts were formed in the reactions of **1c** and **1d**, possibly due to the mismatched configurations of the side chains. The octapeptide **3** incorporated with commonly used acyclic dAAs, Ala(4-Pte) was also synthesized. RCM of (R)- or (S)-Ala(4-Pte)-based peptide **3a**-**3d** under the optimized conditions (A), as well as under the standard conditions for all-hydrocarbon stapling (B)^{7,23} gave a mixture of *E*- and *Z*-isomers with E/Z = 2.1:1-1:2.2 and with moderate yields.

Scheme 1. RCM reaction of octapeptides 1 and 3^a



^{*a*}Isolated yields (0.05 mmol scale) are provided with the *E*/*Z* ratio determined by ¹H NMR. ^{*b*}Including inseparable dimer byproducts. ^{*c*}NMR yield.

Next, we investigated the impact of the substrate on the Eselectivity. In contrast to the decrease of *E*-selectivity (E/Z = ca. 7-12:1) with shorter peptides reported by Grubbs et al.,⁸ our Ac₅c-based peptides retained high E/Z-selectivity even at a pentapeptide level (E/Z = 23-26.1, Table 1, entries 1–3). The octapeptide 5d having L-leucine tripeptide on the C-terminus ensured better *E*-selectivity (E/Z = 40.1, entry 4). The optimized conditions were inappropriate for Val-containing peptide 5e, with a formation of an enormous amount of dimer byproducts and the decreased E/Z of 16:1 (entry 5). The heptapeptide 5f containing Ac₅c^{3OAII} instead of *O*-allyl-L-homoserines in Karle and Balaram's sequence shown in Figure 1A (E/Z = ca. 5:1) improved the *E*-selectivity up to 35:1 (entry 6). Our method is also compatible with the solid-phase synthesis, as we did for peptides 5g and 5h.^{22,23} The best *E*-selectivity was achieved with the **6h** peptide (E/Z = 59:1, entry 8).

Since Ac_5c^{3OAII} in compound **1a** provided a better RCM reaction rate with improved *E*-selectivity than that with commonly used (*S*)-Ala(4-Pte) (**3a**), we compared their crystallographic structures in stapled peptides. The crystallographic structures were successfully obtained for carbocyclic-dAAs-containing peptides **1a** and **2a**, and for acyclic-dAAs-containing peptides **3a**, (*E*)-**4a**, and (*Z*)-**4a** (Figure 2). A superimposed structure of stapled peptides **2a** and **4a** shows that **2a** has larger helical radius than that in **4a**. The crystalline peptides **1a** and **2a** involve four consecutive intramolecular hydrogen bonds $[N(n + 4)H\cdots O=C(n)]$ of the $i \leftarrow i + 4$ type in **1a**, and five consecutive intramolecular hydrogen bonds of the $i \leftarrow i + 4$ type in **2a** (Table S19). These values indicate the existence of an α -helical

Table 1. Effect of peptide sequence in substrate on *E*-selective stapling

$R^{1} - (Xxx)_{m} - N \xrightarrow{S} - Sh$			
Entry	Sequence $[Y = allyl \text{ or } (E)-2-butenyl tether]$	Yield (%)	E/Z^a
1	Boc-Leu-Leu- $(1S,3S)$ -Ac ₅ c ^{3OY} -Leu-Leu-Leu- $(1R,3S)$ -Ac ₅ c ^{3OY} -OMe (5a / 6a)	68	23:1
2	Boc-Leu- $(1S,3S)$ -Ac ₅ c ^{3OY} -Leu-Leu-Leu- $(1R,3S)$ -Ac ₅ c ^{3OY} -OMe (5b/6b)	85	24:1
3	Boc- $(1S,3S)$ -Ac ₅ c ^{3OY} -Leu-Leu-(1 <i>R</i> ,3 <i>S</i>)-Ac ₅ c ^{3OY} -OMe (5 c/6c)	75	26:1
4	Boc- $(1S,3S)$ -Ac ₅ c ^{3OY} -Leu-Leu-Leu- $(1R,3S)$ -Ac ₅ c ^{3OY} -Leu-Leu-Leu-OMe (5d/6d)	98	40:1
5^b	Boc- $(1S,3S)$ -Ac ₅ c ^{3OY} -Val-Val-Val- $(1R,3S)$ -Ac ₅ c ^{3OY} -OMe (5e/6e)	99	16:1
6	Boc-Val- $(1S,3S)$ -Ac ₅ c ^{3OY} -Leu-Aib-Val- $(1R,3S)$ -Ac ₅ c ^{3OY} -Leu-OMe (5f/6f)	78	35:1
$7^{c,d}$	Bz-(1 <i>S</i> ,3 <i>S</i>)-Ac ₅ c ^{3OY} -Leu-Leu-(1 <i>R</i> ,3 <i>S</i>)-Ac ₅ c ^{3OY} -NH ₂ (5 g/6g)	65	23:1
$8^{c,d}$	$\label{eq:Ac-Thr-Phe-(1S,3S)-Ac_5c^{3OY}-Asp-Leu-Leu-(1R,3S)-Ac_5c^{3OY}-Tyr-Gly-Pro-NH_2\ (\mathbf{5h/6h})$	99 ^{e,f}	59:1 ^{<i>f</i>}

^{*a*}Determined by ¹H NMR. ^{*b*}In CH₂Cl₂ at 25 °C. ^{*c*}The result was obtained after cleavage from the resin. ^{*d*}The first-generation Grubbs catalyst was used in CH₂Cl₂ and the reaction was repeated once. ^{*c*}Conversion. ^{*f*}Determined by HPLC.

secondary structure in 1a and 2a. Conversely, crystalline peptide 3a involves four consecutive intramolecular hydrogen bonds of the $i \leftarrow i + 4$ type and one intramolecular hydrogen bond $[N(n + 3)H \cdots O=C(n)]$ of the $i \leftarrow i + 3$ type. Two consecutive intramolecular hydrogen bonds of both the $i \leftarrow i + 4$ type and $i \leftarrow i + 3$ type were observed in (*E*)-4a and (*Z*)-4a. These values indicate the coexistence of the α - and 3_{10} -helical structures. Based on types and numbers of these intramolecular hydrogen bonds, the stapling in 2a increased α -helicity compared to 1a, while the stapling in (*E*)-4a and (*Z*)-4a increased



Figure 2. (A) Crystallographic structures of unstapled peptides (1a, 3a) and stapled peptides [2a, (E)-4a, (Z)-4a]. (B) Superimposed structure of 2a (magenta) and (E)-4a (cyan).

3₁₀-helicity compared to **3a**. Similar trends were observed with the torsion angles, where the average torsion angles of **2a** [avg.(ϕ 1– ϕ 7) = -66.8° and avg.(ψ 1– ψ 7) = -42.0°] are much closer to the ideal values of the right-handed α -helix [ϕ = -63° and ψ = -42°]²⁴ than that of **1a** [avg.(ϕ 1– ϕ 7) = -68.0° and avg.(ψ 1– ψ 7) = -38.8°].²² Conversely, the average torsion angles of peptides **3a**, (*E*)-**4a**, and (*Z*)-**4a** [avg.(ϕ 1– ϕ 7) = -68.0°, -74.5°, and -73.1°; avg.(ψ 1– ψ 7) = -34.7°, -32.3°, and -33.5°, respectively] are much closer to the ideal values of the right-handed 3₁₀-helix [ϕ = -57° and ψ = -30°] rather than those of the α -helix. These results suggest that α -helicity of stapled peptides is better enhanced by cyclic dAAs (Ac₅c^{30All}) than by commonly used acyclic-dAAs [Ala(4-Pte)] in the crystalline state.

Although the detailed reaction mechanism of the *E*-selective side-chain formation is unclear so far, we performed additional experiments illustrated in Scheme 2 to reveal the conditions providing a high *E*-selectivity. The poor *E*/*Z*-selectivities of **4a** was not caused by a macro-ring strain, since the poor *E*-selectivity of 4.2-2.0:1 was also observed for the RCM of **7**, which involves 7-octenyl side chain. Combination of a cyclic dAA, (1S,3S)-Ac₅c^{3OAII}, and an acyclic dAA, (*R*)-Ala(4-Pte), in peptide **9** was not adapted for *E*-selective RCM reaction that resulted in a poor *E*/*Z*-selectivity (*E*/*Z* = 3.4–2.6:1). It was possible to isomerize the olefin geometry of **2a** (*E*/*Z* = 3:1) to a 21:1 ratio under the optimized conditions. These results suggest that high *E*-selectivity is due to the thermodynamic preference of *E*-isomer by the combination of (1S,3S)-Ac₅c^{3OAII} and (1R,3S)-Ac₅c^{3OAII}.

In conclusion, we developed an *E*-selective RCM reaction in α -helical stapled peptide at *i*,*i* + 4 positions. The combination of carbocyclic dAAs, (1*S*,3*S*)-Ac₅c^{3OAII}, and (1*R*,3*S*)-Ac₅c^{3OAII} provided the maximum *E*-selectivity of up to 59:1, while commonly used acyclic (*S*)- and (*R*)-Ala(4-Pte), produced a mixture of *E*- and *Z*-isomers with the *E*/*Z* of 2.1–0.5:1. A variety of peptide sequences with a different amount of residues was detected in the solution-phase and by the solid-phase synthesis. X-ray crystallographic analysis suggested that peptide **2a**, composed **Scheme 2.** Further RCM reactions



^{*a*}Conditions A: The second-generation Grubbs catalyst (20 mol %), toluene (5 mM), 40 °C, 30 min; Conditions B: the first-generation Grubbs catalyst (20 mol %), 1,2-dichloroethane (5 mM), 25 °C, 2 h. ^{*b*}Including inseparable dimer byproducts.

of (1S,3S)-Ac₅c^{3OAII} and (1R,3S)-Ac₅c^{3OAII}, possessed a righthanded α -helical structure, while peptide **4a**, composed of (*S*)-Ala(4-Pte), comprises a mixture of right-handed 3₁₀- and α -helical segment conformations. Further studies including mechanistic investigation and application to helical foldamer organocatalysts,²⁵ as well as biologically active peptides²⁶ are in the progress in our laboratory.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Experimental details, characterization data for all new compounds, and crystallographic analyses. (PDF)

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Notes

The authors declare no competing financial interest.

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