

1 **Episodic knowledge on first and foremost crystalline structural report of**
2 **ethylenediaminetetraacetic acid (EDTA) itself**

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8
9 **ABSTRACT**

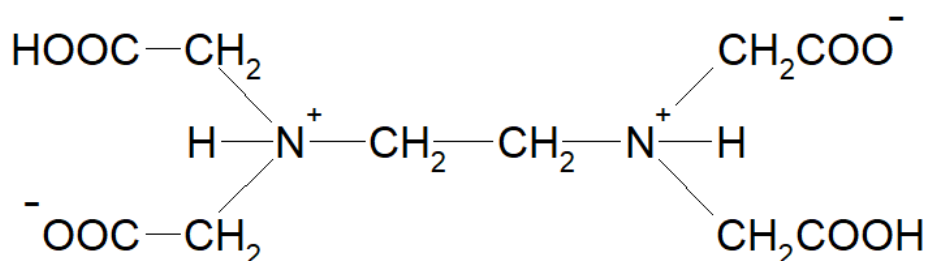
10 An investigation into the precise crystalline structural report of ethylenediaminetetraacetic acid
11 (EDTA), a representative chelator learned in the education of analytical chemistry, was initiated as
12 there appeared to be apparent ambiguity regarding the accurate notation of EDTA as a neutral
13 molecule or a zwitterion. Consequently, the first and foremost article reporting EDTA as
14 zwitterionic form was determined after a difficult confirmation. Additionally, crystallographic data
15 of the zwitterionic EDTA itself and non-chelated salts thereof registered on the CCDC are also
16 summarized herein. This episodic knowledge will be fundamental for not only chemical education
17 but also related fundamental and applied sciences.

18 **Keywords:** chemical analysis, history, polyaminocarboxylic acid, salt.

19
20 **Introduction**

21 Ethylenediaminetetraacetic acid (2,2',2'',2'''-(ethane-1,2-diyl)dinitrilo)tetraacetic acid; EDTA) is a
22 commonly learned chelator in analytical chemistry owing to its ability to form stable metal
23 complexes with four carboxylate groups (R-COO⁻) and two tertiary amino groups (R-NR'R'')
24 thereof. However, the accurate notation of EDTA itself appears somewhat ambiguous. This would
25 be because there was a controversy about the chemical structure of EDTA itself around the middle
26 of the 20th century.^(1,2) The one group interpreted EDTA as a neutral molecule,⁽³⁻⁵⁾ whereas the
27 other group interpreted it as a zwitterion; i.e., HOOCCH₂(⁻OOCCH₂)NH⁺CH₂CH₂NH⁺(CH₂COO⁻
28)CH₂COOH.⁽⁶⁻⁸⁾ Even afterwards, there are still some textbooks of analytical chemistry that

29 describes it as not a zwitterion but a neutral molecule.⁽⁹⁾ Precise understanding and revealing the
30 historical origin of the chemical structure of EDTA itself should be significant not only for
31 chemical educators but also for scientific researchers. Therefore, I have investigated and analyzed
32 the reported crystal structures of both zwitterionic EDTA themselves and its non-chelating salts; 27
33 entries were demonstrated recently.⁽¹⁰⁾ But at that time, the first and foremost article reporting the
34 crystal structure of EDTA had not yet been determined. Eventually, according to a difficult task to
35 confirm, it was found that the first and foremost crystal structure of EDTA was reported with
36 zwitterionic fashion by Lu & Shao in 1961 and 1962 (Fig. 1).^(11,12)



38
39 **Fig. 1** Zwitterionic notation reported by crystal structure of EDTA.^(11,12)

40

41 **Materials and methods**

42 Information as to crystal compounds at Cambridge Crystallographic Data Centre (CCDC) have
43 been searched through the Internet with no fee (<https://www.ccdc.cam.ac.uk/structures/>).

44

45 **Results and Discussion**

46 Hereafter, the complicated episode found by my investigation is demonstrated in detail. The above-
47 mentioned monumental articles (Lu & Shao 1961; 1962) reporting EDTA crystal structures written
48 in Chinese with an abstract in English⁽¹¹⁾ and in Russian⁽¹²⁾ have been reported. However, they have
49 been neglected by almost all researchers reporting the crystal structures of EDTA themselves and
50 their salts of non-chelation after that.⁽¹⁰⁾ Why is this such a regrettable situation? From the web site
51 of the Cambridge Crystallographic Data Centre (CCDC) exhibiting a search result, it is shown that
52 the data by Lu & Shao (1962) was deposited in 1971 as “L.Y.Tsein, S.M.Chen, Scientia Sinica,
53 1962, 9, 496”. I tried to obtain a report through the Internet, but the article was not available online.
54 Thus, I ordered a copy of the article via the University Library. Eventually, I found that as the
55 original article by Lu & Shao (1962) was written in Russian, the bibliography must be translated
56 into English. Unfortunately, the bibliographic notation is incorrect. In Chinese culture, surnames

57 (family names) are traditionally written before ‘first’ (given) names. Usually, when translating
58 Chinese names into English, the surname is often swapped with the given name. However, in the
59 present case, swapping was not done; ‘Tsein’ and ‘Chen’ are indeed parts of their given names.
60 These parts of their given names seemed to be treated as surnames during the translation. Second,
61 although the volume number in the bibliography is noted as “9”, it is also incorrect. In the original
62 article, the volume number is demonstrated as the Roman numeral “XI”; XI means 11, not 9 (=IX)!
63 Thirdly, the page number is incorrect. In the original paper, the page numbers began from 469, not
64 496!! Triple mistakes, in addition to unavailability through the Internet, are making it be very
65 difficult to identify.

66 Subsequently, Cotrait in 1972 reported another data of the crystal structure of EDTA with *R* index
67 written in French.⁽¹³⁾ Cotrait (1972) cited the above-mentioned article as “Tsin, L. Y. & Chen, S. M.
68 (1962). Acta Sci. Sin. p. 469”. The journal name thereof is misleading; “Acta Sci. Sin.” is
69 inaccurate, because ‘Acta’ was unnecessary, and another confusing journal of “Acta Chim. Sin.”
70 also exists. Moreover, the volume number was not included. Later, the further crystal structure of
71 EDTA itself was also reported by Ladd & Povey (1973),⁽¹⁴⁾ citing the aforementioned paper as “Lu,
72 Y. & Shao, M. (1962) Sci. Sin. II, 469”. As you know by now, they mistaken the volume number
73 ‘XI’ for II!!! These incomplete citations are seemed like another reason for losing the worth of the
74 first-reported invaluable knowledge. Here should be also emphasized that none of them never
75 mentioned the actual first and foremost report by Lu & Shao (1961), even though posterior Chinese
76 researchers who rediscovered and deposited the crystal structural data of EDTA itself on the
77 CCDC.⁽¹⁵⁾

78 Further information shall be introduced. I also found additional entries of the non-chelation salts of
79 EDTA such as $[\text{Mg}(\text{H}_2\text{O})_6](\text{EDTA}-2\text{H})$ by Tinnemans in 2023,⁽¹⁶⁾ and two entries of
80 $(\text{HOCH}_2\text{CH}_2\text{NH}_3)_2(\text{EDTA}-2\text{H})$ and $\{(\text{HOCH}_2)_3\text{CNH}_3\}_2(\text{EDTA}-2\text{H})\cdot 3\text{H}_2\text{O}$ by Semenov et al. in
81 2023⁽¹⁷⁾ through this investigational process after reporting an addendum.⁽¹⁸⁾ Accordingly, a total of
82 33 entries related to EDTA themselves and non-chelating salts thereof are shown in Table 1. Here,
83 $(\text{EDTA}+2\text{H})^{2+}$ denotes ethylenediammoniotetra-acetic acid
84 $(\text{HOOCCH}_2)_2\text{NH}^+\text{CH}_2\text{CH}_2\text{NH}^+(\text{CH}_2\text{COOH})_2$; $(\text{EDTA}+\text{H})^+$ denotes ethylenediammonioacetato-tri-
85 acetic acid $(\text{HOOCCH}_2)_2\text{NH}^+\text{CH}_2\text{CH}_2\text{NH}^+(\text{CH}_2\text{COO}^-)\text{CH}_2\text{COOH}$; EDTA denotes
86 ethylenediammoniodiacetato-di-acetic acid $\text{HOOCCH}_2(\text{OOCCH}_2)\text{NH}^+\text{CH}_2\text{CH}_2\text{NH}^+(\text{CH}_2\text{COO}^-$
87 $)\text{CH}_2\text{COOH}$; and $(\text{EDTA}-2\text{H})^{2-}$ denotes ethylenediammoniotetra-acetate ($-$
88 $\text{OOCCH}_2)_2\text{NH}^+\text{CH}_2\text{CH}_2\text{NH}^+(\text{CH}_2\text{COO}^-)_2$, respectively. These existing data of the crystalline
89 structures of zwitterionic EDTA itself and its non-chelational salts are expected to be valuable for
90 both chemical education and sciences afterwards.

Table 1. Non-chelating zwitterionic EDTA molecules and their salts with reported crystal structural data deposited on the CCDC (types of EDTA, entry numbers, chemical formulae, CCDC numbers) sorted by the CCDC Nos.

Type	Entry No.	Chemical formula	CCDC No.
(EDTA+2H) ²⁺	1	EDTA·2HCl·3H ₂ O	657759
	2	EDTA·2HCl·3H ₂ O	1163542
	3	EDTA·H ₂ SO ₄ ·H ₂ O	1300053
(EDTA+H) ⁺	4	EDTA·HBr·H ₂ O	1283049
	5	EDTA·0.39H ₂ O	1101398
EDTA	6	EDTA	1148825
	7	EDTA	1148826
	8	EDTA	1148827
	9	EDTA	1148828
	10	EDTA·(H ₃ O)Na ₂ [Ni(EDTA-2H)]PW ₁₂ O ₄₀ ·5H ₂ O	1433251
	(EDTA-2H) ²⁻	11	[Mn(H ₂ O) ₆](EDTA-2H)
12		N ₂ H ₅ (EDTA-2H)	656243
13		(N ₂ H ₅) ₂ (EDTA-2H)	656246
14		Sr ₂ (EDTA-2H)Cl ₂ ·5H ₂ O	686018
15		Sr ₂ (EDTA-2H)(HCO ₃) ₂ ·4H ₂ O	686020
16		{HC ₃ N ₃ (NH ₂) ₃ } ₂ (EDTA-2H)·2H ₂ O	741062
17		K ₂ (EDTA-2H)·2H ₂ O	838146
18		Na ₂ (EDTA-2H)·2H ₂ O	1020814
19		(NH ₄)(N ₂ H ₅)(EDTA-2H)·NH ₃ ·2H ₂ O	1033967
20		[Mg(H ₂ O) ₆](EDTA-2H)	1103464
21		Rb ₂ (EDTA-2H)·2H ₂ O	1147983
22		K ₂ (EDTA-2H)·2H ₂ O	1148821
23		Ca(EDTA-2H)·2H ₂ O	1166578
24		Na ₂ (EDTA-2H)·2H ₂ O	1171161
25		Na ₂ (EDTA-2H)[Te(OH) ₆]·2H ₂ O	1309261
26		K ₂ (EDTA-2H)·2H ₂ O	1968403
27		Ag ₂ (EDTA-2H)	2006750
28		Na ₂ (EDTA-2H)·2H ₂ O	2051489
29		Ba(EDTA-2H)·4H ₂ O	2054973
30		Sr ₂ (EDTA-2H) ₂ ·6H ₂ O	2054974
31		(HOCH ₂ CH ₂ NH ₃) ₂ (EDTA-2H)	2144221
32	{(HOCH ₂) ₃ CNH ₃ } ₂ (EDTA-2H)·3H ₂ O	2172960	
33	[Mg(H ₂ O) ₆](EDTA-2H)	2241171	

92

93 Conclusions

94 The first and foremost crystalline structural reports exhibiting an exact zwitterionic EDTA itself has
 95 provided valuable insights into its history in chemistry. However, the bibliographic data of this
 96 monumental article reported by Lu & Shao in 1961 was misinterpreted when registered in the
 97 CCDC, and subsequent studies also erroneously cited. This episodic knowledge would be of
 98 significance for not only chemical education but also scientific research in the near future.

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155 **Conflict of interests**

156 The Author declares no conflict of interests.