Graphical Abstract

To create your abstract, type over the instructions in the template box below. Fonts or abstract dimensions should not be changed or altered.

Helical structures of homo-chiral isotope-labeled α -aminoisobutyric acid peptides

Leave this area blank for abstract info.

Atsushi Ueda, Makoto Oba, Yuki Izumi, Yasuhito Sueyoshi, Mitsunobu Doi, Yosuke Demizu, Masaaki Kurihara, and Masakazu Tanaka





Tetrahedron journal homepage: www.elsevier.com



Helical structures of homo-chiral isotope-labeled α -aminoisobutyric acid peptides

Atsushi Ueda^a, Makoto Oba^a, Yuki Izumi^a, Yasuhito Sueyoshi^a, Mitsunobu Doi^b, Yosuke Demizu^c, Masaaki Kurihara^c, and Masakazu Tanaka^a,*

^a Graduate School of Biomedical Sciences, Nagasaki University, 1-14 Bunkyo-machi, Nagasaki 852-8521, Japan

^b Osaka University of Pharmaceutical Sciences, Osaka 569-1094, Japan

^c Division of Organic Chemistry, National Institute of Health Sciences, Tokyo 158-8501, Japan

ARTICLE INFO	ABSTRACT		
Article history:	Type your Abstract text here		
Received	2009 Elsevier Ltd. All rights reserved.		
Received in revised form	The object douterium, and ¹³ C isotone lokaled or aminoisobuturic acids CD. Aib and ¹³ CU. Aib		
Accepted	The child deuterium- and "C-isotope-rabeled o-animoisobutyric actors CD3-Alb and "CH3-Alb		
Available online	transfer catalysts. Homo chiral (S) CD: Aib homopentides up to decemers, were prepared A		
Keywords:	(R)-CD ₃ -Aib polymer and (S) - ¹³ CH ₃ -Aib polymer were also prepared. Conformational studies on		
helix	homopeptides using CD spectra and an X-ray crystallographic analysis revealed that the preferred		
conformation	conformations were 310-helical structures comprising equal amounts of right-handed (P) and left-		
peptide	handed (M) helical-screw structures. The α -carbon chiral centers induced by the D- or ¹³ C-isotope		
Aib	substitution of Aib were incapable of controlling the helical-screw directions of their oligopeptides		
α, α -disubstituted α -amino acid	and short polymers.		

1. Introduction

 α -Aminoisobutyric acid (Aib; α -methylalanine, α, α -dimethylglycine) is an achiral α, α -disubstituted α -amino acid, in which the α -hydrogen atom of alanine is replaced with a methyl substituent.¹ Aib peptides preferentially form 3₁₀-helical structures,² and Aib is often used to design conformational-freedom restricted peptides in the fields of medicinal and organic chemistry.³ Aib is achiral as are its homopeptides; therefore, these helical structures of Aib homopeptides exist as enantiomeric right-handed (*P*) and left-handed (*M*) helical-screw conformers in a ratio of 1 to 1.⁴ Incidentally, if the isotope of an atom is taken into consideration in Aib, for example, the ¹³C-isotope of a methyl substituent at the α -carbon atom accounts for approximately 2% in nature, then approximately 2% of Aib exists as the racemic chiral ¹³CH₃-Aib.

Regarding the relationship between deuterium $D(^{2}H)$ isotope-induced chirality and the polymer helical-screw direction, Green and coworkers reported that the side-chain chirality of isocyanate caused by the replacement of H with the D isotope was capable of controlling its isocyanate polymer (chiral pseudo-peptoid polymer) helical-screw direction.⁵ However, a relationship between an α -carbon chiral center induced by a replacement with an isotope (D or ¹³C) and the helical-screw direction of its peptides has not yet been reported. Thus, we synthesized chiral Aib; CD₃-Aib and ¹³CH₃-Aib, by replacing a methyl substituent of Aib with CD₃⁶ or ¹³CH₃,⁷ and examined the effects of the isotopes (D and ¹³C) on their homopeptide helical-screw directions (Fig. 1).

H ₃ C CH ₃	D ₃ C CH ₃	H ₃ ¹³ C CH ₃
H ₂ N [×] CO ₂ H	H ₂ N [×] CO ₂ H	H₂N ^{, X} CO₂H
Aib	(S) CD ₃ Aib	(<i>R</i>) ⁻¹³ CH ₃ Aib

Fig.1. Chemical structures of Aib, CD₃-Aib, and ¹³CH₃-Aib.

^{*} Corresponding author. Tel. & fax: +81-95-819-2423; e-mail: matanaka@nagasaki-u.ac.jp

2

Tetrahedron

2. Results and Discussion

2.1. Enantioselective synthesis of optically active CD₃-Aib and ¹³CH₃-Aib

We synthesized both enantiomers of optically active (*S*)- and (*R*)-CD₃-Aib using (*R*)- and (*S*)-simplified Maruoka catalysts^{®8}, respectively, as the chiral phase-transfer catalyst (Scheme 1). According to Maruoka's procedure, the alkylation of the aldimine Schiff base **1**, derived from an L-alanine *tert*-butyl (Bu') ester, with CD₃I (2.1 equiv) and CsOH·H₂O (5 equiv) in toluene in the presence of 0.5 mol % (*R*)-simplified Maruoka catalyst[®] {(*R*)-4,4-dibutyl-2,6-bis(3,4,5-trifluorophenyl)-4,5-dihydro-3*H*-dinaphto[7,6,1,2-cde]azepinium bromide} at -40 °C, and a work-up with acidic aqueous citric acid, gave (*S*)-CD₃-Aib-OBu' (*S*)-**2a** in a quantitative yield. In contrast, the reaction of **1** using the (*S*)-simplified Maruoka catalyst[®] produced enantiomeric (*R*)-**2a** in a quantitative yield. We attempted to examine the % ee of **2a** using chiral HPLC; however, the separation of enantiomers did not work well. The % ee of **2a** was established as 90% ee, after the conversion of (*S*)-**2a** into a (*R*)- α -methoxy- α -trifluoromethylphenylacetic acid (MTPA) amide and by measuring the ¹H NMR spectrum.⁹

Both enantiomers of ¹³CH₃-Aib were subsequently synthesized in a similar manner to those of CD₃-Aib. The alkylation of **1** with ¹³CH₃I (1.2 equiv) and CsOH·H₂O (5 equiv) in toluene in the presence of 0.5 mol % (*R*)-simplified Maruoka catalyst[®] at -40 °C and a work-up under acidic conditions gave (*R*)-¹³CH₃-Aib-OBu' **2b** in a quantitative yield.¹⁰ While, the reaction using (*S*)-simplified Maruoka catalyst[®] produced enantiomeric (*S*)-**2b** in quantitative yield. The % ee of (*R*)-**2b** was established after the conversion of **2b** into (*R*)-MTPA amide and by ¹H NMR measurements. The enantiomeric excess of (*R*)-**2b** was *ca.* 82% ee.⁹



Scheme 1. Enantioselective syntheses of (S)-CD₃-Aib and (R)-¹³CH₃-Aib. Abbreviations: p-chlorophenyl (p-Cl-Ph); tetrahydrofuran (THF).

2.2. Preparation of homo-chiral CD₃-Aib homopeptides

We prepared homopeptides, up to the (S)-CD₃-Aib decapeptide and (R)-CD₃-Aib octapeptide, using solution-phase methods, as follows (Scheme 2). The N-terminal-protected Cbz-(S)-CD₃-Aib-OH **3a** was synthesized from (S)-**2a** by benzyloxycarbonyl (Cbz) protection and C-terminal deprotection in 77% yield. The coupling between (S)-**2a** and (S)-**3a** using 1-ethyl-3-(dimethylaminopropyl)carbodiimide hydrochloride (EDC) and 1-hydroxybenzotriazole (HOBt) gave the dipeptide (S)-**4a** in 78% yield. The removal of the protecting groups at the N terminus and C terminus produced the dipeptide amine (S)-**5a** and dipeptide carboxylic acid (S)-**6a**, respectively. Fragment coupling between (S)-**5a** and (S)-**6a** using O-(7-azabenzotriazol-1-yl)-1,1,3,3-tetramethyluronium hexafluorophosphate (HATU) and 1-hydroxy-7-azabenzotriazole (HOAt) gave the tetrapeptide (S)-**7a** in 63% yield. The octapeptide (S)-**10a** (23%) was prepared from the tetrapeptide amine (S)-**11a** and dipeptide carboxylic acid (S)-**6a** in 48% yield.

Enantiomeric (R)-CD₃-Aib homopeptides were prepared in a similar method as that described for (S)-CD₃-Aib homopeptides.



Scheme 2. Preparation of homo-chiral (S)-CD₃-Aib homopeptides.

2.3. Conformational analysis of homo-chiral (S)-CD₃-Aib homopeptides

The (S)-CD₃-Aib tetrapeptide **7a** and octapeptide **10a** provided crystals suitable for an X-ray crystallographic analysis following slow evaporation of the solvent (MeOH/H₂O) at room temperature.

An X-ray crystallographic analysis is incapable of discriminating between isotopes D and H because X-ray diffraction is caused by electrons, and the D and H atoms both have the same electron. The bond lengths of C–H and C–D are slightly different because zeropoint energy differs for the C–H and C–D bonds.¹¹ However, the difference in the two bond lengths is markedly smaller than X-ray analysis precision. Therefore, even if the X-ray crystallographic analysis exclusively showed a one-handed helical structure for (*S*)-CD₃-Aib peptides in the crystal state, we cannot establish the right-handedness or left-handedness of the helix. However, if the crystal structure was solved in the chiral space group and only a one-handed helical structure existed, the helical-screw sense may be controlled by chiral centers at the α -carbon atoms to one-handedness, although its right-handedness or left-handedness cannot be clarified.

The crystal and diffraction parameters of (*S*)-**7a** and (*S*)-**10a** are summarized in Table S1.⁹ Their molecular structures are given in Figures 2 and 3. The relevant backbone and side-chain torsion angles as well as the intra- and intermolecular hydrogen-bond parameters are listed in Tables 1 and 2.^{12,13}

Table 1. Selected torsion angles of homo-chiral (S)-CD₃-Aib homopeptides 7a and 10a.^a

Torsion	(S)-CD ₃ -Aib	(S)- CD ₃ -Aib octamer 10a				
Angle	tetramer 7a	Α	В	С	D	
ω0	175.7	-170.0	-176.1	176.9	167.0	
φ1	-56.9	-59.5	-59.1	54.3	64.0	
ψ1	-36.2	-33.3	-32.0	34.0	24.7	
ω1	-174.5	-178.4	-179.7	176.2	176.8	
φ2	-55.6	-46.2	-51.1	55.5	54.8	
ψ2	-31.2	-36.3	-35.0	30.5	36.4	
ω2	-176.3	-175.4	-176.1	178.6	175.3	
φ3	-59.8	-61.0	-56.0	56.1	54.2	
ψ3	-31.0	-19.7	-23.4	27.9	26.1	
ω3	-172.4	176.1	179.8	180.0	-178.9	

						Tetrahedron
φ4	50.9	-52.5	-56.9	53.3	50.5	
ψ4	41.7	-31.7	-24.4	22.4	25.0	
ω4	177.0	179.9	173.6	-174.7	177.9	
φ5		-51.4	-50.9	47.9	57.0	
ψ5		-26.3	-23.9	35.3	25.8	
ω5		178.1	-177.0	176.2	180.0	
φ6		-51.5	-53.4	45.2	52.1	
ψ6		-28.7	-35.5	41.9	28.0	
ω6		-179.2	-170.8	169.6	178.8	
φ7		-54.3	-69.0	61.6	56.9	
ψ7		-28.0	-15.6	19.9	26.7	
ω7		-176.3	175.1	179.3	170.6	
φ8		54.0	56.0	-51.9	-51.0	
ψ8		42.2	40.1	-44.2	-42.7	
ω8		174.0	167.8	-167.9	-176.3	

 $^{\rm a}$ The number of amino acid residues begins at the N terminus of the peptide chain.

Table 2. Intra- and intermolecular H-bond parameters for (S)-CD₃-Aib peptides 7a and 10a.

Peptide	Donor D–H	Acceptor A	Distance [Å] D…A	Angle [°] D– H…A	Symmetry operations		
$Cbz-\{(S)-CD_3-Aib\}_4-OBu^t$ (7a)							
	N ₃ -H	O_0	3.04	153	x,y,z		
	N ₄ -H	O_1	2.99	159	x,y,z		
	N ₁ -H	O _M '	2.83	140	2-x,1-y, -z		
	N ₂ -H	O _{4'}	3.03	121	-1/2+x, 3/2-y,		
					-1/2+z		
	O _M -H	O _{2'}	2.72	175	3/2-x, -1/2+y,		
					1/2 - Z		
$Cbz-\{(S)$	$-CD_3-Aib\}_{8}$	OBu^t (10a)					
$A\left(P ight)$	N _{3a} -H	O_{0a}	3.15	157	x,y,z		
	N _{4a} -H	O_{1a}	3.01	160	x,y,z		
	N _{5a} -H	O_{2a}	3.04	164	x,y,z		
	N _{6a} -H	O_{3a}	2.92	165	x,y,z		
	N _{7a} -H	O_{4a}	3.05	165	x,y,z		
	N_{8a} -H	O _{5a}	3.00	166	x,y,z		
B(P)	N _{3b} -H	O_{0b}	2.93	151	x,y,z		
	N _{4b} -H	O _{1b}	2.95	162	x,y,z		
	N _{5b} -H	O_{2b}	3.03	164	x,y,z		
	N _{6b} -H	O _{3b}	2.90	167	x,y,z		
	N _{7b} -H	O_{4b}	3.10	164	x,y,z		
	N _{8b} -H	O _{5b}	3.02	161	x,y,z		
$C\left(M ight)$	N _{3c} -H	O_{0c}	2.95	155	x,y,z		
	N _{4c} -H	O _{1c}	2.92	159	x,y,z		
	N _{5c} -H	O_{2c}	2.98	165	x,y,z		
	N _{6c} -H	O_{3c}	2.86	163	x,y,z		
	N _{7c} -H	O_{4c}	3.03	162	x,y,z		
	N _{8c} -H	O _{5c}	3.07	163	x,y,z		
$D\left(M ight)$	N _{3d} -H	O_{0d}	3.10	154	x,y,z		
	N _{4d} -H	O_{1d}	2.88	161	x,y,z		
	N _{5d} -H	O_{2d}	2.98	170	x,y,z		
	N _{6d} -H	O_{3d}	2.99	165	x,y,z		
	N_{7d} -H	O_{4d}	2.92	167	x,y,z		
	N _{8d} -H	O _{5d}	2.96	171	x,y,z		
	N _{1a} -H	O _{7a} ,	2.90	160	x, -1+y,z		

N _{1b} -H	O _{7b'}	2.81	176	x, -1+y,z
N_{1c} -H	O _{7c'}	2.80	173	x, 1+y,z
N_{1d} -H	O_{7d}	2.87	163	x, 1+y,z
O_{M4} -H ^a	O_{4c}	3.02	176	x,y,z
O_{M1} -H ^a	O _{4b'}	2.86	177	x,y,1+z
O _{M2} -H ^a	O _{M1}	2.87	153	x,y,z

^a O_M: O atom of MeOH.

The (*S*)-CD₃-Aib tetrapeptide **7a** was solved in the space group P_{2i}/n to form a 3₁₀-helical structure in the asymmetric unit. The space group P_{2i}/n is centrosymmetric; and therefore, right-handed (*P*) and left-handed (*M*) helical enantiomers existed in the crystal. Figure 2 shows a right-handed (*P*) 3₁₀-helical structure, in which reversal of the C-terminal torsion angles occurred, *i.e.*, the signs of the ϕ and ψ torsion angles of the C-terminal residue (residue 4) were opposite to those of the preceding residues (1–3). This phenomenon is frequently observed in 3₁₀-helical Aib and related peptides.¹⁴ The average values of the torsion angles ϕ and ψ of residues (1–3) were –57.4°, –32.9°. The intramolecular hydrogen bonds of the $i \leftarrow i+3$ type (3₁₀-helix) were found between H–N(3) and the C(0)=O(0) oxygen atom of the Cbz-group, and between H–N(4) and the C(1)=O(1). In the packing mode, the (*P*) conformer is connected to the symmetry-related (-1/2+x, 3/2-y, -1/2+z) (*M*) conformer by one intermolecular hydrogen bond [H–N(2)…O(4')].



Fig. 2. A right-handed (*P*) 3_{10} -helical structure of the (*S*)-CD₃-Aib tetrapeptide **7a** by an X-ray crystallographic analysis. An enantiomeric left-handed (*M*) 3_{10} -helix also existed. Views (a) along the helical axis, and (b) perpendicular to the helical axis.

The structure of the (*S*)-octapeptide **10a** was solved in the space group *P1*. Two right-handed (*P*) 3_{10} -helices and two left-handed (*M*) 3_{10} -helices existed in the asymmetric unit, along with methanol molecules. The space group *P1* is chiral and non-centrosymmetric; therefore, the right-handed (*P*) helices and left-handed (*M*) helices are not enantiomeric, but diastereomeric. This result may be due to solvent molecules disturbing the symmetry of the helical molecule arrangement. Reversal of the C-terminal torsion angles occurred in conformers *A*–*D*, *i.e.*, the signs of the ϕ and ψ torsion angles of the C-terminal residue (residue 8) were opposite to those of the preceding residues (1–7). The average values of the torsion angles ϕ and ψ of residues (1–7) were *A*: –53.8°, –29.1°; *B*: –56.6°, –27.1°; *C*: 53.4°, 30.3°; *D*: 55.6°, 27.5°. Conformers *A*–*D* showed six intramolecular hydrogen bonds of the *i*←*i*+3 type (3₁₀-helix) between H–N(3) and the C(0)=O(0) oxygen atom of the Cbz-group, between H–N(4) and C(1)=O(1), between H–N(5) and C(2)=O(2), between H–N(6) and C(3)=O(3), between H–N(7) and C(4)=O(4), and between H–N(8) and C(5)=O(5). In the packing mode, an intermolecular hydrogen bond formed between the H-N(1) peptide donor and C7'=O7' of a symmetry-related conformer (*A* and *B*: x, –1+y, z) or (*C* and *D*: x, 1+y, z). Thus, the chains of intermolecularly hydrogen-bonded right-handed (*P*) 3_{10} -helices; …*A*···*A*···*A*··· A··· A··· and …*B*···*B*···, and left-handed (*M*) 3_{10} -helices; …*C*···*C*··· C··· C··· and …*D*···*D*···*D*···, were formed in a head-to-tail alignment.



Tetrahedron

6

b)

Fig 3. Right-handed (P) and left-handed (M) 3₁₀-helical structures of the (S)-CD₃-Aib octapeptide 10a by an X-ray crystallographic analysis. Views (a) perpendicular to the helical axes, and (b) along the helical axes.

The X-ray crystallographic structure of the achiral Aib octapeptide *p*-BrBz-(Aib)₈-OBu' (*p*-BrBz: *p*-bromobenzoyl) has already been solved in the centrosymmetric space group *P*-1, and enantiomeric (*P*) and (*M*) 3_{10} -helices existed, together with three methanol molecules.^{4e} These differences in the X-ray crystallographic results between the achiral Aib octapeptide^{4e} and our chiral CD₃-Aib octapeptide **10a** may be attributed to the difference in the N-terminal protecting groups of peptides, but not to the effects of CD₃-induced chiral centers. These X-ray crystallographic results suggest that chiral centers at the α -carbons of (*S*)-CD₃-Aib peptides are incapable of controlling their helical-screw sense to one-handedness up to an octapeptide length.

The FT-IR absorption spectra of Cbz-{(*S*)-CD₃-Aib}_n-OBu' {n = 4 (**7a**), 8 (**10a**), 10 (**12a**)} showed weak bands in the 3420–3440 cm⁻¹ region, which were assigned to solvated free peptide NH groups, and strong bands at the 3350–3370 cm⁻¹ region to peptide NH groups with N–H···O=C intramolecular hydrogen bonds of different strengths.⁹ These FT-IR spectra are almost the same as those of the reported Aib homopeptides, which formed 3₁₀-helices.⁴ The CD spectra of the (*S*)-CD₃-Aib octa- and decapeptides **10a** and **12a** in 2,2,2-trifluoroethanol (TFE) solution (0.10 mM) did not show maxima (208 and 222 nm) for the one-handed helical structure (Fig. 4). These results also suggest that right-handed (*P*) and left-handed (*M*) 3₁₀-helices exist in equal amounts.¹⁵

Toniolo and coauthors reported that right-handed and left-handed helices of achiral Aib homopeptides rapidly interconverted in solution at room temperature.¹⁶ Therefore, the right-handed and left-handed helices of chiral (*S*)-CD₃-Aib homopeptides also might rapidly interconvert in solution at room temperature.



Fig. 4. CD spectra of (S)-CD₃-Aib homopeptides Cbz-{(S)-CD₃-Aib}_n-OBu^t {(a) n = 8 (10a), (b) n = 10 (12a)} in TFE solution (0.10 mM).

2.4. (R)-CD₃-Aib and (S)-¹³CH₃-Aib polymer syntheses based on N-carboxy anhydride methods and their CD spectra

We attempted to prepare a (*R*)-CD₃-Aib polymer and (*S*)-¹³CH₃-Aib polymer based on *N*-carboxy anhydride method. According to Deming's procedure,¹⁷ the Cbz-protected CD₃-Aib (*R*)-**3a** was converted to the *N*-carboxy anhydride (*R*)-**13a** in 91% yield by a treatment with SOCl₂ in THF at 65 °C. Furthermore, the Cbz-protected ¹³CH₃-Aib (*S*)-**3b**, which was prepared from (*S*)-**2b**, was similarly converted to the *N*-carboxy anhydride (*S*)-**13b** in 69% yield.

Initiators for polymerization: *n*-hexyl amine, triethylamine (Et₃N), and 1,8-diazabicyclo[5.4.0]undec-7-ene (DBU), were examined using the achiral Aib *N*-carboxy anhydride **13**.^{18,19} The MALDI-ToF-MS spectra of products showed molecular ion peaks of m/z 1230

(14 mers) by *n*-hexyl amine, m/z 2594 (30 mers) by Et₃N, and m/z 3443 (40 mers) by DBU. The use of the strongest base DBU as an initiator in *N*,*N*-dimethylformamide (DMF) gave the highest polymerization degree. Thus, we polymerized the isotope-labeled chiral Aib *N*-carboxy anhydrides **13a**,**b** under these reaction conditions.



Scheme 3. Preparation of (R)-CD₃-Aib and (S)-¹³CH₃-Aib N-carboxy anhydrides and their polymerization.

The MALDI-ToF-MS spectrum of the (*R*)-CD₃-Aib polymer **14a** showed larger molecular ion peaks than 42 mers {3744.3: (M+Na)⁺}, and these peaks appeared at regular intervals of m/z 88, which corresponded to the monomer unit of $-HN-C_3H_3D_3-CO-$. On the other hand, the MALDI-ToF-MS spectrum of the (*S*)-¹³CH₃-Aib polymer **14b** showed larger molecular ion peaks than 30 mers {2635.5: (M+K)⁺}, and the interval of these peaks was m/z 86, corresponding to the monomer unit of $-HN-^{13}CC_2H_6-CO-$ (Fig. 5).

a)



Fig. 5. MALDI-ToF-MS spectra of (a) the (R)-CD₃-Aib polymer and (b) (S)-¹³CH₃-Aib polymer.

The CD spectra of the (*R*)-CD₃-Aib polymer **14a** and (*S*)- 13 CH₃-Aib polymer **14b** in TFE solution did not show any positive or negative maxima for the one-handed helical structure. These CD spectra suggest that these polymers form right-handed and left-handed helical structures in almost equal amounts, and the control of the helical-screw sense of polymers is not possible at these polymer lengths (Fig. 6).



Fig. 6. CD spectra of (a) the (R)-CD₃-Aib polymer 14a (0.3 mg in TFE 1.0 mL) and (b) (S)-¹³CH₃-Aib polymer 14b (0.4 mg in TFE 1.0 mL).

Helical-screw control by chiral centers due to isotope substitutions were different between Green's pseudo-peptoid polymers and our Aib oligomers. In Green's study,⁵ isocyanate polymers showed an excess of one-handed helical-screw structures in the CD spectra. In our study, neither (*R*)-CD₃-Aib peptides nor (*S*)-¹³CH₃-Aib peptides had a bias for the one-handed helical-screw direction detected in the CD spectra; however, they had right-handed and left-handed helical-screw directions in equal amounts. The most prominent difference between isocyanate polymers and Aib polymers is molecular weight, and thus, the numbers of existing chiral centers. The average molecular weight (M_w) of Green's isocyanate polymers was 870,000; therefore, *ca.* 6,800 unit polymers and nearly 6,800 chiral centers existed. However, our Aib polymers were 30-50 mers, with only 30-50 chiral centers existing.

3. Conclusion

Optically active isotope-labeled Aib **2a** and **2b** were synthesized using simplified Maruoka catalysts[®]. The (*S*)-CD₃-Aib homopeptides, up to decamers, were prepared by peptide-fragment coupling using a solution-phase method. Homo-chiral CD₃-Aib and ¹³CH₃-Aib polymers, up to ca. 30-40 mers, were also prepared by the polymerization of *N*-carboxy anhydrides. The conformational analysis of these peptides demonstrated that right-handed and left-handed helical structures existed in equal amounts, and 30-40 chiral centers caused by D and ¹³C isotope substitutions were incapable of affecting the helical-screw directions of their oligopeptides. Isotope-labeled drugs have recently attracted the attention of medicinal chemists.²⁰ Therefore, these results demonstrate that isotope-labeled drugs have negligible effects on conformational changes.

4. Experimental Section

4.1. General

Optical rotations $[\alpha]_D$ were measured using a 1.0-dm cell. Circular dichroism spectra (CD) were measured using a 1.0-mm path length cell. Infrared spectra (IR) were recorded for conventional measurements (neat or KBr), and the solution (CDCl₃) method was performed using the 0.1-mm path length of a NaCl cell. ¹H NMR spectra were determined at 400 MHz at room temperature. EI(MS) and HRMS(FAB) spectra were taken in the dual-focusing sector field mode, while HRMS(ESI), HRMS(DART), and MALDI-MS spectra were measured in the ToF mode.

4.2. Synthesis of isotope-labeled amino acids

4.2.1. tert-*Butyl* 2-(4-Chlorobenzylideneamino)propionate (1).²¹ A mixture of H-Ala-OBu^t·HCl (30.0 g, 165 mmol), NaHCO₃ (10 g) in H₂O (200 mL), and CH₂Cl₂ (200 mL) was stirred at room temperature for 2 h. The organic layer was separated, *p*-chlorobenzaldehyde (13.3 g, 14.6 mmol) and anhydrous MgSO₄ (16 g) were then added, and the whole mixture was stirred at room temperature for 4 days. After filtration, the filtrate was evaporated and diluted with diethyl ether. The ethereal solution was washed with brine and H₂O, and then dried over MgSO₄. After removal of the solvent, the residue was recrystallized from *n*-pentane to give the imine **1** (20.1 g, 79%) as colorless crystals: mp 25–27 °C; IR (neat) v 2982, 1732, 1645, 1217, 1155, 1090, 770, 747 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 8.26 (s, 1H), 7.72 (d, *J* = 8.3 Hz, 2H), 7.38 (d, *J* = 8.3 Hz, 2H), 4.04 (q, *J* = 6.8 Hz, 1H), 1.48 (d, *J* = 6.8 Hz, 3H), 1.47 (s, 9H); HRMS(DART): [M+H]⁺, found 268.1111. C₁₄H₁₉ClNO₂ requires 268.1104.

4.2.2. tert-*Butyl* (S)-2-*Amino*-2-(*trideuteriomethyl*)propionate [H-(S)-CD₃-Aib-OBu^t; (S)-2a]. CD₃-I (2.00 mL, 32.0 mmol) was added dropwise to a stirred solution of **1** (4.00 g, 14.9 mmol), (*R*)-simplified Maruoka catalyst[®] (0.5 mol %, 56 mg, 0.075 mmol), and CsOH·H₂O (12.5 g, 74.7 mmol) in toluene (80 mL) at -40 °C, and the solution was stirred for 24 h. The solution was diluted with water, and extracted with CHCl₃. Evaporation of the solvent afforded the oily residue, which was dissolved in THF (250 mL) and 0.5 M aqueous citric acid

solution (250 mL), and the solution was stirred for 4 h. After removal of THF, the residual solution was washed with *n*-hexane, basified with Na₂CO₃ (*ca.* 44 g), extracted with diethyl ether, and dried over MgSO₄. Removal of the solvent gave an oily residue, which was distilled under reduced pressure using Kugelrohr apparatus to leave **2a** (2.45 g, quantitative) as a colorless oil: bp 100–110 °C/300 mbar (oil bath temp.); $[\alpha]^{30}_{D} \pm 0.0$ (*c* 1.31, CHCl₃); IR (neat) v 3422 (br), 2980, 2878, 1717, 1636, 1275 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.74 (br s, 2H), 1.46 (s, 9H), 1.29 (s, 3H); HRMS(DART): [M+H]⁺, found 163.1532. C₈H₁₅D₃NO₂ requires 163.1526. Enantiomeric (*R*)-**2a** was synthesized using (*S*)-simplified Maruoka catalyst[®], and racemic (±)-**2a** was synthesized using tetrabutyl ammonium iodide instead of Maruoka catalyst[®].

4.2.3. Determination of enantiomeric excess of 2a. A mixture of amine 4a (9 mg, 0.057 mmol), (*R*)-MTPA (16 mg, 0.068 mmol), and EDC (13.6 mg, 0.071 mmol) in CH₂Cl₂ (0.5 mL) was stirred at 50 °C for 4 days. After removal of the solvent, the residue was purified by short column chromatography on silica gel to give an MTPA amide. Enantiomeric excess of 2a was determined by the ¹H NMR spectrum of the corresponding MTPA amide.⁹

4.2.4. tert-*Butyl* (*R*)-2-*Amino*-2-(13 C-*methyl*)propionate [*H*-(R)- 13 CH₃-Aib-OBu^t; (R)-2b].⁷ ¹³CH₃-I (56 µL, 0.90 mmol) was added dropwise to a stirred solution of **1** (200 mg, 0.75 mmol), (*R*)-simplified Maruoka catalyst[®] (0.5 mol %, 4.2 mg, 0.006 mmol), and CsOH·H₂O (66 mg, 4.40 mmol) in toluene (4 mL) at -40 °C, and the solution was stirred for 24 h. The solution was diluted with water, and extracted with CHCl₃. Evaporation of the solvent afforded the oily residue, which was dissolved in THF (13 mL) and 0.5 M aqueous citric acid solution (13 mL), and the solution was stirred for 4 h. After removal of THF, the residual solution was washed with *n*-hexane, basified with Na₂CO₃, extracted with ether, and then dried over MgSO₄. Removal of the solvent gave an oily residue, which was distilled under reduced pressure using Kugelrohr apparatus to leave (*R*)-**2b** (120 mg, quantitative) as a colorless oil: bp 100–110 °C/380 mbar (oil bath temp.); [α]²⁹_D ±0.0 (*c* 1.10, CHCl₃); IR (neat) v 3372 (br), 1724, 1369, 1142 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.66 (br s, 2H), 1.46 (s, 9H), 1.30 (d, ⁴J_{CH} = 4.4 Hz, 3H), 1.29 (d, ¹J_{CH} = 128 Hz, 3H); HRMS(DART): [M+H]⁺, found 161.1329. C₇¹³CH₁₈NO₂ requires 161.1371. Enantiomeric (*S*)-**2b** was synthesized using (*S*)-simplified Maruoka catalyst[®], and racemic (±)-**2b** was synthesized using tetrabutyl ammonium iodide instead of Maruoka catalyst[®].

4.2.5. Determination of enantiomeric excess of **2b**. A mixture of amine **4b** (9 mg, 0.057 mmol), (*R*)-MTPA (16 mg, 0.068 mmol), and EDC (13.6 mg, 0.071 mmol) in CH₂Cl₂ (0.5 mL) was stirred at 50 °C for 4 days. After removal of the solvent, the residue was purified by column chromatography on silica gel to give an MTPA amide. Enantiomeric excess of **2b** was determined by the ¹H NMR spectrum of the corresponding MTPA amide.⁹

4.3. Preparation of (S)-CD₃-Aib homopeptides

4.3.1. (S)-2-(*Benzyloxycarbonyl)amino*-2-(*trideuteriomethyl*)propionic acid [Cbz-{(S)-CD₃-Aib}-OH; (S)-**3a**]. A mixture of (S)-**2a** (2.00 g, 12.3 mmol), Na₂CO₃ (1.57 g, 14.8 mmol), and benzyloxycarbonyl chloride (Cbz-Cl; 2.60 mL, 18.4 mmol) in acetone-water (1 : 1; 50 mL) was stirred at room temperature for 4 days. After removal of acetone, the aqueous solution was extracted with ethyl acetate (EtOAc), and dried over Na₂SO₄. After removal of the solvent, the residue was purified by column chromatography on silica gel (20% EtOAc in *n*-hexane) to give Cbz-protected amino ester (3.00 g, 83%) as colorless crystals. Cbz-{(S)-CD₃-Aib}-OBu': mp 60–62 °C; $[\alpha]^{20}$ _D ±0.0 (*c* 0.99, CHCl₃); IR (KBr) v 3372 (br), 2976, 1711 (br), 1518, 1302, 1260, 1146 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.30–7.36 (m, 5H), 5.46 (br s, 1H), 5.08 (s, 2H), 1.51 (s, 3H), 1.43 (s, 9H); HRMS(DART): [M+H]⁺, found 297.1913. C₁₆H₂₁D₃NO₄ requires 297.1894. A mixture of Cbz-{(S)-CD₃-Aib}-OBu' (1.38 g, 4.66 mmol) and anisole (0.9 mL, 8.30 mmol) in 2,2,2-trifluoroacetic acid (TFA; 9.2 mL) was stirred at room temperature for 3 h. Removal of the solvent afforded a solid, which was purified by short column chromatography on silica gel (20% EtOAc in *n*-hexane) to give (S)-**3a** (1.11 g, 99%) as colorless crystals: mp 70–71 °C; $[\alpha]^{28}_{D} \pm 0.0$ (*c* 1.00, CHCl₃); IR (KBr) v 3333 (br), 2988, 2945, 2907, 1750 (br), 1653, 1543, cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 9.20 (br s, 1H), 7.30–7.40 (m, 5H), 5.39 (br s, 1H), 5.10 (s, 2H), 1.57 (s, 3H); HRMS(DART): [M+H]⁺, found 241.1251. C₁₂H₁₃D₃NO₄ requires 241.1268.

4.3.2. Cbz-[(S)- CD_3 - $Aib]_2$ - OBu^t [(S)-4a]. A mixture of amine (S)-2a (353 mg, 2.17 mmol), carboxylic acid (S)-3a (447 mg, 1.86 mmol), EDC (432 mg, 2.25 mmol), and HOBt (308 mg, 2.28 mmol) was stirred at 60 °C for 3 days. After removal of the solvent, the residue was diluted with EtOAc, washed with 3% aqueous HCl, 5% aqueous NaHCO₃, and then dried over Na₂SO₄. After removal of the solvent, the white solid was purified by column chromatography on silica gel (20% EtOAc in *n*-hexane) to produce a terminally-protected dipeptide (*S*)-4a (558 mg, 78%) as colorless crystals: mp 136–137 °C; $[\alpha]^{31}_{D} \pm 0.0$ (*c* 1.00, CHCl₃); IR (KBr) v 3406, 3294 (br), 2978, 1721, 1659, 1535 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.31–7.36 (m, 5H), 6.91 (br s, 1H), 5.36 (br s, 1H), 5.09 (s, 2H), 1.52 (s, 3H), 1.49 (s, 3H), 1.45 (s, 9H); HRMS(DART): [M+H]⁺, found 385.2626. C₂₀H₂₅D₆N₂O₅ requires 385.2610.

4.3.3. $H_{[(S)-CD_3-Aib]_2-OBu^t}[(S)-5a]$. A mixture of (S)-4a (951 mg, 2.47 mmol), and 10% Pd-C (760 mg) in MeOH (70 mL) was rigorously stirred under a H₂ atmosphere at room temperature for 13 h. After filtration, the filtrate was evaporated to give a crude amine (S)-5a (619 mg, quantitative), which was used for the next reaction without purification. ¹H NMR (400 MHz, CDCl₃) δ 8.09 (br s, 1H), 1.73 (br s, 2H), 1.51 (s, 3H), 1.45 (s, 9H), 1.34 (s, 3H); HRMS(DART): [M+H]⁺, found 251.2256. C₁₂H₁₉D₆N₂O₃ requires 251.2242.

4.3.4. Cbz-[(S)- CD_3 - $Aib]_2$ -OH [(S)-6a]. A mixture of (S)-4a (113 mg, 0.294 mmol) and anisole (60 µL) in trifluoroacetic acid (0.6 mL) was stirred at room temperature for 4 h. Removal of the solvent afforded a white solid, which was used without purification.

4.3.5. $Cb_{z-[}(S)-CD_3-Aib]_4-OBu^t[(S)-7a]$. A mixture of (S)-5a (242 mg, 0.968 mmol), (S)-6a (273 mg, 0.832 mmol), HATU (597 mg, 1.57 mmol), HOAt (217 mg, 1.59 mmol), and diisopropylethylamine ([†]Pr₂EtN) (0.40 mL) in MeCN (5 mL) was stirred at 50 °C for 5 days under an Ar atmosphere. After removal of the solvent, the residue was diluted with EtOAc, washed with 3% aqueous HCl, 5% aqueous NaHCO₃, and then dried over Na₂SO₄. Removal of the solvent gave a white solid, which was purified by column chromatography on silica gel (40% EtOAc in *n*-hexane) to produce a terminally-protected tetrapeptide (S)-7a (293 mg, 63%) as colorless crystals: mp 184–185 °C; $[\alpha]^{26}_{D} \pm 0.0$ (*c* 1.00, CHCl₃); IR (CDCl₃): v 3426, 3358 (br), 2982, 1717, 1674, 1519 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.30–

7.40 (m, 6H), 7.24 (br s, 1H), 6.59 (br s, 1H), 5.51 (br s, 1H), 5.09 (s, 2H), 1.44 (s, 3H), 1.43 (s, 3H), 1.43 (s, 9H), 1.39 (s, 3H), 1.26 (s, 3H); HRMS(DART): $[M+H]^+$, found 561.4002. $C_{28}H_{33}D_{12}N_4O_7$ requires 561.4041.

4.3.6. H-[(S)- CD_3 -Aib]₄- OBu^t [(S)-8a]. Amine (S)-8a was prepared from (S)-7a in a manner similar to that described for the preparation of (S)-5a, and used for the next reaction without purification.

4.3.7. Cb_2 -[(S)- CD_3 - Aib_{4} -OH [(S)-9a]. Carboxylic acid (S)-9a was prepared from (S)-7a in a manner similar to that described for the preparation of (S)-6a, and used for the next reaction without purification.

4.3.8. Cbz-[(S)- CD_3 -Aib]₈- OBu^t [(S)-**10a**]. Terminally-protected octapeptide (S)-**10a** was prepared by coupling between (S)-**8a** and (S)-**9a** in a manner similar to that described for the preparation of (S)-**7a**. 23%; colorless crystals; mp 241–243 °C; $[\alpha]^{22}_{D} \pm 0.0$ (*c* 1.00, CHCl₃); IR (CDCl₃) v 3316 (br), 2984, 1713, 1661, 1530 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.69 (br s, 1H), 7.66–7.67 (br s, 2H), 7.58 (br s, 1H), 7.52 (br s, 1H), 7.44 (br s, 1H), 7.28–7.42 (m, 5H), 6.64 (br s, 1H), 5.58 (s, 1H), 5.11 (s, 2H), 1.48 (s, 3H), 1.46 (s, 6H), 1.45 (s, 15H), 1.44 (s, 3H), 1.38 (s, 3H), 1.30 (s, 3H); HRMS(FAB): [M]⁺, found 912.6824. C₄₄H₄₈D₂₄N₈O₁₁ requires 912.6827.

4.3.9. $H_{-[(S)-CD_3-Aib]_8-OBu^t}[(S)-11a]$. Amine (S)-11a was prepared from (S)-10a in a manner similar to that described for the preparation of (S)-5a, and used for the next reaction without purification.

4.3.10. *Cbz-[(S)-CD₃-Aib]*₁₀-*OBu^t* [(S)-**12a**]. Terminally-protected decapeptide (S)-**12a** was prepared by coupling between (S)-**11a** and (S)-**6a** in a manner similar to that described for the preparation of (S)-**7a**. 48%; colorless crystals; mp >250 °C; $[\alpha]^{23}_{D} \pm 0.0$ (*c* 1.00, CHCl₃); IR (CDCl₃) v 3321 (br), 2984, 1715, 1661, 1530 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.70 (br s, 1H), 7.66 (br s, 1H), 7.65 (br s, 1H), 7.64 (br s, 1H), 7.61 (br s, 1H), 7.53 (br s, 1H), 7.39 (br s, 2H), 7.35–7.38 (m, 5H), 6.43 (br s, 1H), 5.63 (br s, 1H), 5.12 (s, 2H), 1.53 (s, 3H), 1.47–1.48 (m, 21H), 1.44 (s, 12H), 1.31 (s, 3H); HRMS(FAB): [M]⁺, found 1088.8254. C₅₂H₅₆D₃₀N₁₀O₁₃ requires 1088.8259.

4.4. Synthesis of N-carboxy anhydrides of chiral CD₃-Aib and ¹³CH₃-Aib

4.4.1. (R)-4-trideuteriomethyl-4-methyloxazoline-2,5-dione [(R)-13a]. Thionyl chloride (SOCl₂; 40 µL, 0.55 mmol) was added to the stirred solution of (*R*)-3a (32.0 mg, 0.133 mmol) in THF (1 mL) at 65 °C under an Ar atmosphere. After being stirred for 6 h, the solution was evaporated, and the residue was recrystallized from *n*-hexane/EtOAc (5/1) to give (*R*)-13a (16.0 mg, 91%) as colorless crystals: mp 95–96 °C; $[\alpha]^{28}_{D} \pm 0.0$ (*c* 0.50, MeOH); IR (KBr) v 3568 (br), 2974, 2862, 1856, 1782 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 6.37 (br, 1H), 1.58 (s, 3H); MS(EI): *m/z* 132 [M]⁺.

4.4.2. (S)-2-(*Benzyloxycarbonyl)amino*-2-(13 C-methyl)propionic acid [Cbz-{(S)- 13 CH₃-Aib}-OH; (S)-**3b**]. A mixture of (S)-**2b** (107 mg, 0.670 mmol), Na₂CO₃ (87 mg, 0.82 mmol), and benzyloxycarbonyl chloride (Cbz-Cl; 142 µL, 1.0 mmol) in acetone-water (1 : 1; 50 mL) was stirred at room temperature for 2 days. After removal of acetone, the aqueous solution was extracted with EtOAc, and the extract was dried over Na₂SO₄. After removal of the solvent, the residue was purified by column chromatography on silica gel (20 % EtOAc in *n*-hexane) to give Cbz-protected amino ester (159 mg, 80%) as colorless crystals. Cbz-{(S)- 13 CH₃-Aib}-OBu': mp 58–59 °C; [α] 31 D ±0.0 (*c* 1.00, CHCl₃); IR (KBr) v 3372 (br), 2976, 1711 (br), 1520, 1306, 1260, 1150 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.25–7.36 (m, 5H), 5.46 (br s, 1H), 5.08 (s, 2H), 1.51 (d, $^{1}J_{CH} = 129$ Hz, 3H), 1.51 (d, $^{4}J_{CH} = 4.4$ Hz, 3H), 1.43 (s, 9H); HRMS(DART): [M+H]⁺, found 295.1739. C₁₅ 13 CH₂₄N₁O₄ requires 295.1739. A mixture of Cbz-{(S)- 13 CH₃-Aib}-OBu' (137 mg, 0.466 mmol) and anisole (0.1 mL, 9.2 mmol) in TFA (1 mL) was stirred at room temperature for 4 h. Removal of the solvent afforded a white solid, which was purified by short column chromatography on silica gel (20% EtOAc in *n*-hexane) to give (S)-**3b** (87.5 mg, 79%) as colorless crystals: mp 64–65 °C; [α]²⁹D ±0.0 (*c* 0.71, CHCl₃); IR (KBr) v 3333 (br), 2990 (br), 1724, 1694, 1535 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.30–7.38 (m, 5H), 5.32 (br s, 1H), 5.11 (s, 2H), 3.10 (br, 1H), 1.59 (d, $^{1}J_{CH} = 130$ Hz, 3H), 1.59 (d, $^{4}J_{CH} = 4.2$ Hz, 3H); HRMS(DART): [M+H]⁺, found 239.1128. C₁₁¹³CH₁₆N₁₀Q₄ requires 239.1112.

4.4.3. (S)-4-¹³C-methyl-4-methyloxazoline-2,5-dione [(R)-**13b**]. Thionyl chloride (SOCl₂; 40 µL, 0.55 mmol) was added to the stirred solution of (S)-**3b** (32.0 mg, 0.134 mmol) in THF (1 mL) at 65 °C under an Ar atmosphere. After being stirred for 7 h, the solution was evaporated, and the residue was recrystallized from *n*-hexane/EtOAc (5/1) to give (S)-**13b** (12.0 mg, 69%) as colorless crystals: mp 95–96 °C; $[\alpha]^{28}_{D} \pm 0.0$ (*c* 0.24, MeOH); IR (KBr) v 3356 (br), 3175 (br), 2978, 2870, 1855, 1786 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.84 (br s, 1H), 1.58 (d, ¹J_{CH} = 131 Hz, 3H), 1.58 (d, ⁴J_{CH} = 4.1 Hz, 3H); MS(EI): *m*/z 130 [M]⁺.

4.5. Polymerization Procedure of N-carboxy anhydrides

4.5.1. A solution of the chiral Aib *N*-carboxy anhydride (*S*)-**13** (0.160 mmol) and DBU (1.14 μ L) in DMF (1 mL) was stirred at room temperature overnight under an Ar atmosphere. Evaporation and concentration *in vacuo* gave a polymer.

Acknowledgments

This work was supported in part by a Grant-in-Aid for Scientific Research on Innovative Areas 'Advanced Molecular Transformations by Organocatalysts' (26105745) from the Ministry of Education, Culture, Sports Science, and Technology, and Special Coordination Funds for Promoting Science and Technology from the Japan Science and Technology Agency. (S)- and (R)-simplified Maruoka catalysts[®] were provided by courtesy of NAGASE & Co., Ltd.

Supplementary data

Supplementary data (crystal and diffraction parameters, ¹H NMR spectra of (*R*)-MTPA amides, and FT-IR absorption spectra) associated with this manuscript may be found in the online version, at xxxxxxxx.

References and notes

- (a) Heimgartner, H. Angew. Chem. Int. Ed. 1991, 30, 238–264. (b) Toniolo, C.; Crisma, M.; Formaggio, F.; Valle, G.; Cavicchioni, G.; Précigoux, G.; Aubry, A.; Kamphuis, J. Biopolymers 1993, 33, 1061–1072. (c) Benedetti, E. Biopolymers (Pept. Sci.) 1996, 40, 3–44. (d) Wysong, C. L.; Yokum, T. S.; McLaughlin, M. L.; Hammer, R. P. Chemtech 1997, 26–33. (e) Toniolo, C.; Crisma, M.; Formaggio, F.; Peggion, C. Biopolymers (Pept. Sci.) 2001, 90, 396–419. (f) Tanaka, M. Chem. Pharm. Bull. 2007, 55, 349–358. (g) Crisma, C.; De Zotti, M.; Formaggio, F.; Peggion, C.; Moretto, A.; Toniolo, C. J. Pept. Sci. 2015, 21, 148–177. (h) Crisma, M.; Toniolo, C. Biopolymers (Pept. Sci.) 2015, 104, 46–64.
- 2. Karle, I. L.; Balaram, P. Biochemistry 1990, 29, 6747-6756.
- (a) Rossi, P.; Felluga, F.; Tecilla, P.; Formaggio, F.; Crisma, M.; Toniolo, C.; Scrimin, P. J. Am. Chem. Soc., 1999, 121, 6948–6949. (b) Sissi, C.; Rossi, P.; Felluga, F.; Formaggio, F.; Palumbo, M.; Tecilla, P.; Toniolo, C.; Scrimin, P. J. Am. Chem. Soc., 2001, 123, 3169–3170. (c) Takagi, R.; Shiraki, A.; Manabe, T.; Kojima, S.; Ohkata, K. Chem. Lett. 2000, 366–367. (d) Caporale, A.; Gesiot, L.; Sturlese, M.; Wittelsberger, A.; Mammi, S.; Peggion, E. Amino Acids 2012, 43, 207–218. (e) Guarise, C.; Shinde, S.; Kibler, K.; Ghirlanda, G.; Prins, L. J.; Scrimin, P. Tetrahedron 2012, 68, 4346–4352. (f) Mazumder, A.; Maiti, A.; Roy, K.; Roy, S. ACS Chemical Biology 2012, 7, 1084–1094. (g) Akagawa, K.; Takigawa, S.; Sho, I.; Umezawa, R.; Kudo, K. Org. Lett. 2013, 15, 4964–4967. (h) Akagawa, K.; Kudo, K. Angew. Chem. Int. Ed. 2012, 51, 12786– 12789. (i) Akagawa, K.; Sakai, N.; Kudo, K. Angew. Chem. Int. Ed. 2015, 54, 1822–1826.
- (a) Shamala, N.; Nagaraj, R.; Balaram, P. J. Chem. Soc., Chem. Comm., 1978, 996–997.
 (b) Benedetti, E.; Bavoso, A.; Di Blasio, B.; Pavone, V.; Pedone, C.; Crisma, M.; Bonora, G. M.; Toniolo, C. J. Am. Chem. Soc. 1982, 104, 2437–2444.
 (c) Wilkening, R. R.; Stevens, E. S.; Bonora, G. M.; Toniolo, C. J. Am. Chem. Soc. 1983, 105, 2560–2561.
 (d) Toniolo, C.; Bonora, G. M.; Barone, V.; Bavoso, A.; Benedetti, E.; Di Blasio, B.; Grimaldi, P.; Lelj, F.; Pavone, V.; Pedone, C. Macromolecules 1985, 18, 895–902.
 (e) Toniolo, C.; Bonora, G. M.; Bavoso, A.; Benedetti, E.; Di Blasio, B.; Pavone, V.; Pedone, C. Macromolecules 1986, 19, 472–479.
 (f) Toniolo, C.; Crisma, C.; Bonora, G. M.; Benedetti, E.; Di Blasio, B.; Pavone, V.; Pedone, C.; Santini, A. Biopolymers 1991, 31, 129–138.
 (g) Vlassi, M.; Brueckner, H.; Kokkinidis, M. Acta Cryst. 1993, B49, 560–564.
 (h) Gessmann, R.; Bruckner, H.; Petratos, K. J. Pept. Sci. 2003, 9, 753–762.
- (a) Green, M. M.; Andreola, C.; Munoz, B.; Reidy, M. P. J. Am. Chem. Soc. 1988, 110, 4063–4065. (b) Lifson, S.; Andreola, C.; Peterson, N. C.; Green, M. M. J. Am. Chem. Soc. 1989, 111, 8850–8858. (c) Gu, H.; Nakamura, Y.; Sato, T.; Teramoto, A.; Green, M. M.; Andreola, C.; Peterson, N. C.; Lifson, S. Macromolecules 1995, 28, 1016–1024. (d) Green, M. M.; Park, J.-W.; Sato, T.; Teramoto, A.; Lifson, S.; Selinger, R. L. B.; Selinger, J. V. Angew. Chem. Int. Ed. 1999, 38, 3138–3154.
- 6. (a) Weihrauch, T.; Leibfritz, D. *Liebigs Ann. Chem.* **1985**, 1917–1921. (b) Kawasaki, T.; Shimizu, M.; Nishiyama, D.; Ito, M.; Ozawa, H.; Soai, K. *Chem Commun.* **2009**, 4396–4398.
- (a) Sola, J.; Fletcher, S. P.; Castellanos, A.; Clayden, J. Angew. Chem. Int. Ed. 2010, 49, 6836–6839. (b) Fletcher, S. P.; Sola, J.; Holt, D.; Brown, R. A.; Clayden, J. Beilstein J. Org. Chem. 2011, 7, 1304–1309.
- 8. (a) Ooi, T.; Kameda, M.; Maruoka, K. J. Am. Chem. Soc. 1999, 121, 6519–6520. (b) Kitamura, M.; Shirakawa, S.; Maruoka, K. Angew. Chem. Int. Ed. 2005, 44, 1549–1551.
- 9. See supplementary data.
- 10. The priority rule of the CIP nomenclature is $N > {}^{13}C > {}^{12}C$, and, thus, *R/S* descriptors are different between CD₃-Aib and ${}^{13}CH_3$ -Aib. The absolute configurations of products were assumed based on the reaction mechanisms of the simplified Maruoka catalyst[®].
- 11. Carpenter, B. K. Determination of Organic Reaction Mechanisms, John Wiley: New York, 1984.
- (a) Sheldrick, G. M. Program for Crystal Structure Refinement (SHELXL 97): University of Göttingen: Göttingen, 1997; (b) Beurskens, P. T.; Admiraal, G.; Beurskens, G.; Bosman, W. P.; de Gelder, R.; Israel, R.; Smits, J. M. M. The DIRDIF-99 Program System, Technical Report of the Crystallography Laboratory, University of Nijmegen: The Netherlands, 1994.
- 13. CCDC-1482529 (**7a**) and 1482530 (**10a**) contain the supplementary crystallographic data for this manuscript. These data may be obtained free of charge from The Cambridge Crystallographic Data Centre via: www.ccdc.cam.ac.uk/data_request/cif.
- 14. (a) Shamala, N.; Nagaraj, R.; Balaram, P. J. Chem. Soc., Chem. Commun. **1978**, 996–997. (b) Benedetti, E.; Bavoso, A.; Di Blasio, B.; Pavone, V.; Pedone, C.; Crisma, M.; Bonora, G. M.; Toniolo, C. J. Am. Chem. Soc. **1982**, 104, 2437–2444.
- (a) Toniolo, C.; Polese, A.; Formaggio, F.; Crisma, M.; Kamphuis, J. *J. Am. Chem. Soc.* **1996**, *118*, 2744–2745. (b) Yoder, G.; Polese, A.; Silva, R. A. G. D.; Formaggio, F.; Crisma, M.; Broxterman, Q. B.; Kamphuis, J.; Toniolo, C.; Keiderling, T. A. *J. Am. Chem. Soc.* **1997**, *119*, 10278–10285. (c) Pengo, P.; Pasquato, L.; Moro, S.; Brigo, A.; Fogolari, F.; Broxterman, Q. B.; Kaptein, B.; Scrimin, P. Angew. Chem. Int. Ed. **2003**, *42*, 3388–3392.
- (a) Hummel, R.-P.; Toniolo, C.; Jung, G. Angew. Chem. Int. Ed. 1987, 26, 1150–1152.
 (b) Crisma, M.; De Zotti, M.; Formaggio, F.; Peggion, C.; Moretto, A.; Toniolo, C. J. Pept. Sci. 2015, 21, 148–177.
- 17. Kramer, J. R.; Deming, T. J. Biomacromolecules 2010, 11, 3668-3672.
- 18. (a) Szwarc, M. Adv. Polym. Sci. 1965, 4, 1–65. (b) Sekiguchi, H. Pure Appl. Chem. 1981, 53, 1689–1714. (c) Deming, T. J. J. Polym. Sci. Part A: Polym. Chem. 2000, 38, 3011–3018.
- (a) Malcolm, B. R. *Biopolymers* 1983, 22, 319–322. (b) Dwivedi, A. M.; Krimm, S.Malcolm, B. R. *Biopolymers* 1984, 23, 2025–2065. (c) Malcolm, B. R.; Walkinshaw, M. D. *Biopolymers* 1986, 25, 607–625.
- 20. (a) Sanderson, K. Nature 2009, 458, 269. (b) Sanderson, K. Nature 2009, 458, 817.
- 21. Shirakawa, S.; Yamamoto, K.; Liu, K.; Maruoka, K. Org. Synth. 2013, 90, 121-129.