Microstructures and Thermoelectric Properties of Sintered Misfit-Layered Cobalt Oxide.

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Abstract

Misfit-layered cobalt oxide Ca₃Co₄O₉ is considered to be a prospective material for thermoelectric conversion. The thermoelectric properties are anisotropic owing to its anisotropic crystal structure. The crystal has preferred thermoelectric properties along *a*-*b* plane. Therefore, the thermoelectric properties are improved and controlled by degree of orientation of the sintered sample. In the present work, Sr doped misfit cobalt oxides Ca_{2.7}Sr_{0.3}Co₄O₉ was prepared by solid-phase reaction, sequent uniaxial compression molding and sintering at 1173 K. Seebeck coefficient α , electrical resistivity ρ , and dimensionless figure of merit *ZT* were measured with change of the compression pressure in the uniaxial molding. α , ρ , and *ZT* as functions of the degree of orientation and the relative density are experimentally clarified and explained by the calculations of the complex model.

Key words: Seebeck coefficient; dimensionless figure of merit; degree of orientation; Lotgering factor; Harman method

Introduction

The misfit cobalt oxides Ca₃Co₄O₉ are good representatives of thermoelectric materials because of the high dimensionless figure of merit ZT and superior thermal stability at high temperatures. The unit cell is monoclinic and consists of alternate stacks of a triple rocksalt Ca₂CoO₃ layer (RS) and a CdI₂-type CoO₂ layer along the *c*-axis. The two types of layers have the same *a*, *c* and β lattice parameters but different b_{CoO2} and b_{RS} parameters.^{1,2} The CoO₂ layer are considered responsible for electrical transport, while rocksalt layers are for charge reserve that supplies charge carriers into CoO₂ layers. The thermoelectric properties are anisotropic owing to this anisotropic crystal structure and it has preferred thermoelectric properties along a-b plane.³⁻⁶ Therefore, the thermoelectric properties of the sintered sample are improved by the incease in degree of orientation, which is dependent on the compression pressure in the uniaxial molding. On the other hand, the addition of Sr to $Ca_3Co_4O_9$ can effectively improve the thermoelectric properties owing to the large solution limit of Sr in the RS layer.⁷⁻¹³ In the present work, Sr doped misfit cobalt oxides Ca_{2.7}Sr_{0.3}Co₄O₉ was prepared by solid-phase reaction, sequent uniaxial compression molding and sintering at 1173 K. The effect of the compression pressure in the uniaxial molding on the microstructures and thermoelectric properties was investigated by X-ray diffraction (XRD) and by measurments of the Seebeck coefficient α , electrical conductivity ρ , and ZT. The dependences of α and ρ on the degree of orientation f and relative density d were experimentally revealed. Furthermore, the dependences were calculated and explained by the complex model, in which the sintered sample is composed of many Ca₃Co₄O₉ elements with different orientations and void elements.

Experimental procedure

 $CaCO_3$, Co_3O_4 , and $SrCO_3$ powders were weighed according to the proportions of the $Ca_{2.7}Sr_{0.3}Co_4O_9$ composition. The powders were thoroughly mixed with acetone in an agate mortar, heated at 1123 K for 20 h in an alumina crucible in atmosphere, and then slowly cooled to room temperature. The products were ground into powder (~100 µm), molded at uniaxial compression of 150-900 MPa, and then sintered at

1173 K for 20 h in atmosphere to fabricate pellet samples. The compression direction was perpendicular to the basal plane of the pellets.

The Seebeck coefficient α_{exp} was determined by measuring the voltage at both ends of the pellets, which is caused by the difference in temperature. Electrical resistivity ρ_{exp} was measured using the four-point probe method at room temperature. The dimensionless figure of merit *ZT* was measured by the Harman method.¹⁴ The measurements of the thermoelectric properties were performed parallel to the basal plane of the pellet samples in atmosphere. For the XRD samples, the pellets were ground into powder (~50 µm). The XRD patterns were obtained by using a Rigaku Rint-2200VL and Cu K_{\alpha} radiation for the powder and pellet samples. The degrees of orientation *f* of the pellet samples were estimated by using the XRD peak intensities as Lotgering factor.¹⁵ The relative densities *d* of the pellet samples were obtained by Archimedian method.

Results and discussion

Figure 1 (a) shows α_{exp} (\bigcirc) and ρ_{exp} (\bullet) measured at room temperature as a function of the compression pressure in the uniaxial molding. α_{exp} was positive, indicating p-type conductivity. α_{exp} slightly increases and ρ_{exp} decreases with the increase in the compression pressure. These results are similar to the previous report [4] although the preparing process is different. These may be attributed to the increase in *f* and *d*. *f* increases with the compression pressure because the *a-b* planes of Ca₃Co₄O₉ are aligned parallel to the basal plane of the pellet as shown in the SEM images and the pole figures of Ref. [4]. *d* increases with the compressure because of extinction of the void in the pellet. Fig. 1 (b) shows *ZT* (\bullet) and the power factor *PF* (\bigcirc) at room temperature as a function of the compression pressure. They increase with the compression pressure through the increase in α_{exp} and the decrease in ρ_{exp} .

Figure 2 shows *ZT* as a function of the measured temperature for various compression pressures. *ZT* increases with the measured temperature. The trend that *ZT* increases with the compression pressure holds for high temperature.

Figures 3 (a) shows the X-ray diffraction patterns of $Ca_{2.7}Sr_{0.3}Co_4O_9$ for the powder sample and the pellet samples molded at 150-900 MPa. Most diffraction peaks are indexed as the misfit cobalt oxide $Ca_3Co_4O_9$ phase. The peak intensities other than 00*l* peaks decrease with the increase in the compression pressure because *a-b* planes of the cleaved $Ca_3Co_4O_9$ particles are aligned parallel to the basal plane of the pellet. *f* is quantitatively estimated as the Lotgering factor:¹⁵

$$f=\frac{P-P_0}{1-P_0},$$

$$P = \frac{\sum_{l} I(00l)}{\sum_{hkl} I(hkl)},$$

where I(00l) and I(hkl) indicate the integrated intensities of 00l and hkl peaks, respectively. P_0 and P correspond to the powder and the pellet samples, respectively. Fig. 3 (b) shows f and d as a function of the compression pressure. f increases with the increase in the compression pressure. d also increases because of the extinction of the void.

 α_{exp} and ρ_{exp} measured at room temperature are shown as \bigcirc and \bigcirc (and smoothed solid lines) in Fig. 4 (a) and (b) as a function of *f* and *d*, respectively. α_{exp} slightly increases and ρ_{exp} decreases with the increase in *f* and *d* owing to the alignment of the cleaved Ca₃Co₄O₉ particles and/or the extinction of the voids.

In order to clarify the effect of *f* and *d* on α_{exp} and ρ_{exp} , the calculations based on complex model^{16,17} were performed. In this model, the sample was assumed to be the complex composed of A and B elements as shown in Fig. 5. When the volume fractions of the A and B elements are defined to be C_A and C_B $(C_A+C_B=1)$, the Seebeck coefficient α_{comp} and the electrical resistivity ρ_{comp} along the measured direction *y* for the complex were formulated by the combination of serial and parallel alignments of the A and B elements as^{16,17}

$$\rho_{\text{comp}} = \rho_A (1 - K) + \frac{\rho_A \rho_B}{\rho_B \left(1 - \frac{C_B}{K}\right) + \rho_A c_B/K} K, \quad (1)$$

$$\alpha_{\text{comp}} = \alpha_A (1 - K) + \frac{\alpha_A \rho_B \left(1 - \frac{C_B}{K}\right) + \alpha_B \rho_A C_B/K}{\rho_B \left(1 - \frac{C_B}{K}\right) + \rho_A c_B/K} K.^{16,17} \quad (2)$$

Here, α_A , α_B , ρ_A and ρ_B are the Seebeck coefficients and the electrical resistivities for A and B elements, respectively. *K* is the ratio of the *x*-*z* layers including the B element to the all *x*-*z* layers perpendicular to *y*-axis. *C*_B/*K* is the averaged fraction of B element within the *x*-*z* layer including both of A and B elements. In Eqs. (1) and (2), the first terms of the right hand sides indicate the electrical resistivity and the Seebeck coefficient for the *x*-*z* layers including only A element, respectively. The second terms of the right hand sides indicate those for the *x*-*z* layers including both of A and B elements. The equations indicate that two kinds of layers are serially aligned along the measured direction *y*. If the B element is assumed to be randomly distributed in the cuboid pellet with the width *w*, length *l*, and height *h*, *K* is estimated as $C_B^{\frac{l}{w+l+h}}$. This is because the boundary conditions are satisfied as K = 1 for *h*, $w \gg 1$ and $l \ll 1$ and as $K = C_B$ for *h*, $w \ll 1$, and $l \gg 1$.

In the present case, we assumed the sintered pellet samples as the complex of three kinds of elements: the void and two kinds of oxide elements whose *a-b* planes are aligned parallel to and perpendicular to the measured direction (*y*). At the first step, the sintered pellets are simply assumed to be the complex composed of the oxide and the void elements. The volume fractions of the oxide and the void elements are defined to be $C_{ox}=d$ and $C_v=1-d$, respectively. By substituting the actual pellet size (*w*=*l*=8.86 mm and *h*=1.0 mm) for *K* in the Eqs. (1) and (2), the electrical resistivity and the Seebeck coefficient for the pellet are estimated as

$$\rho_{\rm cal} = \rho_{\rm ox} [1 - (1 - d)^{0.4733}] + \frac{\rho_{\rm ox}}{1 - (1 - d)^{0.5267}} (1 - d)^{0.4733}, \tag{3}$$

 $\alpha_{\rm cal} = \alpha_{\rm ox},$ (4)

where α_A , α_B , ρ_A , and ρ_B in Eqs. (1) and (2) were substituted by α_{ox} , $\alpha_v=0$, ρ_{ox} , and $\rho_v=\infty$, respectively. At the second step, the oxide element is simply assumed to be the complex composed of two kinds of elements whose *a-b* planes are aligned parallel to and perpendicular to the measured direction *y*. The volume fractions of the parallel and the perpendicular elements are approximately assumed to be C_{\parallel} =(2+f)/3 and $C \perp = (1-f)/3$, respectively, by using the degree of orientation *f*. By the assumption of the cubic oxide element, the electrical resistivity and the Seebeck coefficient for the oxide elements are estimated as

$$\begin{aligned} \rho_{\text{ox}} &= \rho_{\parallel} \left[1 - \left(\frac{1-f}{3}\right)^{1/3} \right] + \frac{\rho_{\perp} \rho_{\parallel}}{\rho_{\perp} \left[1 - \left(\frac{1-f}{3}\right)^{2/3} \right] + \rho_{\parallel} \left(\frac{1-f}{3}\right)^{2/3}} \left(\frac{1-f}{3}\right)^{1/3}, \end{aligned} \tag{5} \\ \alpha_{\text{ox}} &= \alpha_{\parallel} \left[1 - \left(\frac{1-f}{3}\right)^{1/3} \right] + \frac{\alpha_{\parallel} \rho_{\perp} \left[1 - \left(\frac{1-f}{3}\right)^{2/3} \right] + \alpha_{\perp} \rho_{\parallel} \left(\frac{1-f}{3}\right)^{2/3}}{\rho_{\perp} \left[1 - \left(\frac{1-f}{3}\right)^{2/3} \right] + \rho_{\parallel} \left(\frac{1-f}{3}\right)^{2/3}} \left(\frac{1-f}{3}\right)^{1/3}, \end{aligned} \tag{6}$$

where α_A , α_B , ρ_A , and ρ_B in Eqs. (1) and (2) were substituted by α_{\parallel} , α_{\perp} , ρ_{\parallel} , and ρ_{\perp} , respectively. ρ_{cal} and α_{cal} can be calculated when *f* and *d* in Fig. 3 (b) are substituted for Eqs. (3)-(6) and $\rho_{\parallel} = 5.7 \text{ m}\Omega\text{cm}$, $\rho_{\perp} = 22.7 \text{ m}\Omega\text{cm}$, $\alpha_{\parallel} = 113.9 \text{ µV/K}$, and $\alpha_{\perp} = 53.0 \text{ µV/K}$, which were reported for non-doped Ca₃Co₄O₉ single crystal at room temperature in Ref. 6, are substituted for them. ρ_{cal} and α_{cal} are shown as \diamondsuit and \blacklozenge (and smoothed dashed lines) in Figs. 4(a) and (b). ρ_{cal} decreases and α_{cal} slightly increases with the increase in *f* and *d* in analogy with ρ_{exp} and α_{exp} . ρ_{exp} is slightly larger than ρ_{cal} over the present range of *d* and *f*. This is because the grain boundaries or the crystal defects are not considered in the calculation. α_{exp} is slightly larger than α_{cal} over the present range of *d* and *f*. This is because the effects of Sr addition on the increase in α are not considered in the calculation. Although the absolute values slightly deviate, the calculations can reproduce the relative change of ρ_{exp} and α_{exp} with the change of *d* and *f*.

This suggests that the complex model composed of the void, the parallel, and perpendicular oxide elements well explains the dependence of the thermoelectric properties for the sintered $Ca_3Co_4O_9$ pellet on the compression pressure.

Conclusions

Sr doped misfit cobalt oxide Ca_{2.7}Sr_{0.3}Co₄O₉ was prepared by solid-phase reaction, sequent uniaxial compression molding and sintering at 1173 K for 20 h. The thermoelectric properties were measured with change of the compression pressure of 150-900 MPa in the uniaxial molding. The XRD measurements and the Archimedian experiments revealed that the degree of orientation *f* and the relative density *d* of the sintered sample increase with the increase in the compression pressure. α_{exp} , *ZT*, and *PF* increase and

 ρ_{exp} decreases with the increase in the compression pressure owing to the increase in *f* and *d*. α_{cal} and ρ_{cal} as functions of *f* and *d* are calculated by the complex model, which composed of the void element, two kinds of oxide elements with different orientations. The calculations can reproduce the relative change of ρ_{exp} and α_{exp} with the change of *d* and *f*.

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Figure captions

Fig. 1. Seebeck coefficient α_{exp} (\bigcirc) and the electrical resistivity ρ_{exp} (\bigcirc) (a), and dimensionless figure of merit *ZT* (\bigcirc) and power factor *PF* (\bigcirc) (b) measured at room temperature as a function of the compression pressure in the uniaxial molding.

Fig. 2. Dimensionless figure of merit ZT as a function of the measured temperature for various compression pressures in uniaxial molding.

Fig. 3. X-ray diffraction patterns of Ca_{2.7}Sr_{0.3}Co₄O₉ for the powder sample and the pellet samples molded at 150-900 MPa (a) and degree of orientation $f(\bullet)$ and relative density $d(\bigcirc)$ of sintered pellet as a function of the compression pressure (b).

Fig. 4. Seebeck coefficient α_{exp} (\bigcirc) and electrical resistivity ρ_{exp} (\blacklozenge) measured at room temperature as a function of the degree of orientation *f* (a) and of the relative density *d* (b) of the pellet samples (solid lines). Seebeck coefficient α_{cal} (\diamondsuit) and electrical resistivity ρ_{cal} (\blacklozenge) calculated by complex model as a function of the degree of orientation *f* (a) and the relative density *d* (b) of the pellet samples (dashed lines). Fig. 5. Scheme of complex model composed of A and B elements, which are randomly distributed. Seebeck coefficient α_{cal} and electrical resistivity ρ_{cal} are calculated along *y*-axis by the complex model.





Fig. 2



Fig. 3



Fig. 4



Fig. 5